

Dating of young ground water with CFCs, SF₆, ³H, and ³H/³He: Principles and Examples

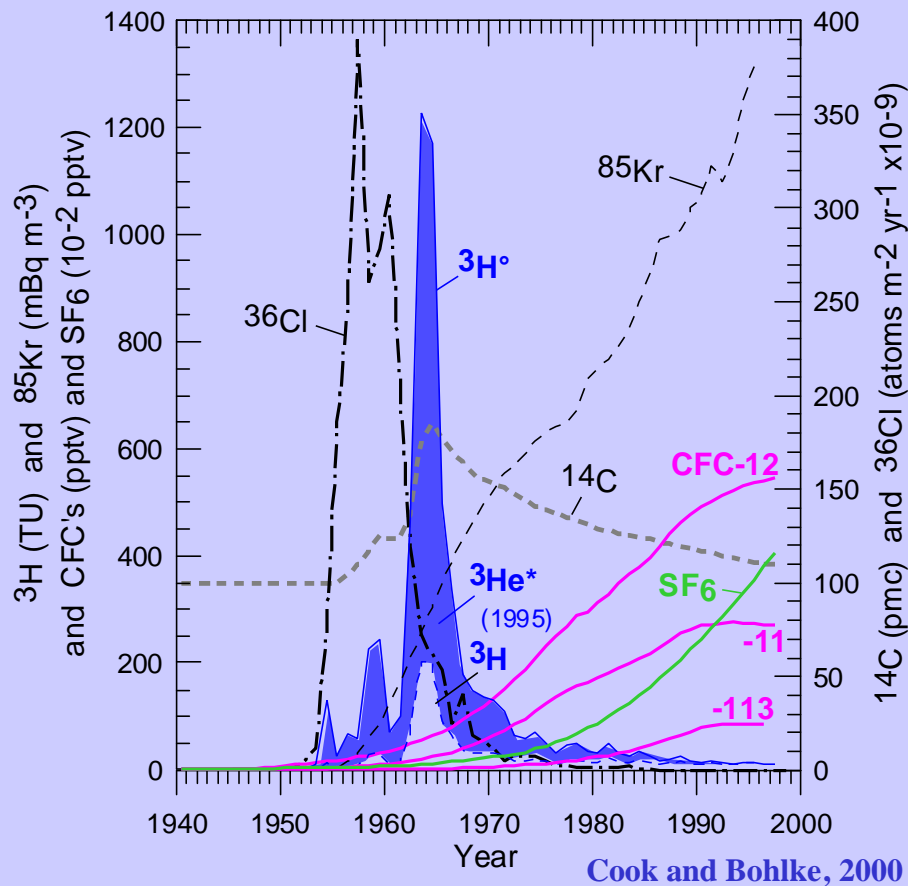
L.N. Plummer, USGS, Reston, VA

U.S. Geological Survey, Office of Water Quality
Forensic Hydrology Workshop, Annapolis, MD
September 1, 2004



Selected Environmental Tracers 0-50 Year Timescale

tracer1b.grf (JK Böhlke) 4/22/03



- ^3H , $^3\text{H}/^3\text{He}$
- ^{85}Kr
- CFC-11, CFC-12, CFC-113
- SF₆
- Event Markers: ^3H , ^{36}Cl , ^{14}C
- Age: time elapsed since recharge

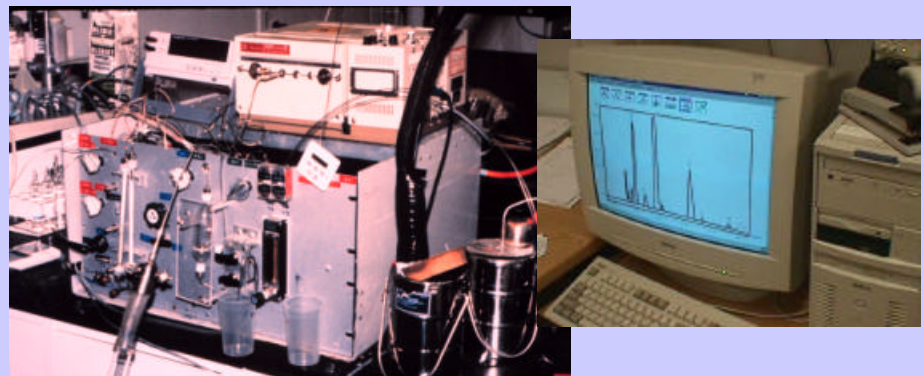
Why measure environmental tracers in ground water?

- Estimate fractions of young water and the mean age of the young fraction in mixtures.
- Evaluate vulnerability to contamination.
- Estimate recharge rates.
- Calibrate models of groundwater flow.
- Estimate rates of geochemical and microbiological processes.
- Date historical records of contaminant loading to aquifers.
- Estimate remediation times.



Approach to “Dating” Young GW

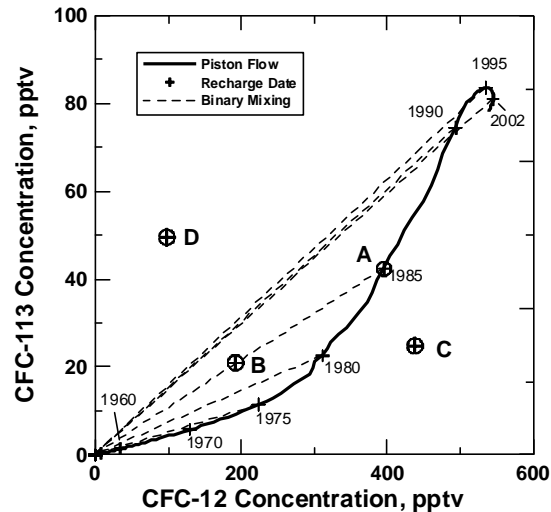
- Collect water samples without contacting air.
- Minimize mixing effects by sampling monitoring wells with narrow screens.
- Analyze with high precision for CFCs, SF₆, ³H, ³H/³He, and others (multi-tracer approach).
- **Age interpretation.** Evaluate multiple tracer data in context of models of groundwater flow.
- **Age is model dependent.**



Age Interpretation

- Comparison of simulated and observed tracer concentrations.
Lumped-parameter models. (1) multiple tracers from the source, (2) a time series from the source, or (3) multiple tracers from multiple sources in the system. *Choose a model based on hydrogeology.*
- **Tracer plots.** Method of comparing simulated and observed multiple tracer data; recognizing cases of possible piston flow and binary mixing (dilution); elimination of some mixing models.
- **Flow-model calibration** and simulation of age. Use of age information to calibrate the model. (Reilly et al., 1994; Inverse modeling; UCODE)
- All “ages”, regardless of method, are **model dependent**.
- **Contradictions?** What’s wrong with the tracers? What’s wrong with the model?

Lumped Parameter Models



IAEA Guidebook on the Use of
Chlorofluorocarbons in Hydrology
2004 Edition



INTERNATIONAL ATOMIC ENERGY AGENCY, VIENNA, 2004

QCFC

<http://www.iaea.org/programmes/rial/pci/isotopehydrology/>

USGS-CFC2004

<http://water.usgs.gov/lab/cfc>

TRACERMODEL1

J.K. Böhlke, is a Microsoft Excel spreadsheet program.

CFC

IAEA Isotope Hydrology Laboratory Excel based program.

FLOW (FLOWPC)

Lumped-parameter models, FORTRAN, Maloszewski and Zuber (1996).

TRACER

EXCEL workbook (Bayari, 2002)

<http://www.iamg.org/CGEditor/index.htm>

LUMPED

Visual Basic (Ozyurt and Bayari, 2003)

<http://www.iamg.org/CGEditor/index.htm>

<http://www.sukimyasilab.hacettepe.edu.tr/english/software.shtm>

BOXMODEL

EXCEL workbook (Zoellmann, Kinzelbach & Aeschbach-Hertig)

http://www.baum.ethz.ch/iHW/boxmodel_en.html



<http://water.usgs.gov/lab/cfc>

USGS <!--USERINHOUSE-->(Internal Access)<!--/USERINHOUSE--> - CFC Lab | Reston, VA



The Reston Chlorofluorocarbon Laboratory (Internal Access)

Site Map

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Welcome

See extensive bibliography



**Dating of
young
groundwater
with
CFCs, SF₆, ³H,
& ³H/³He**
PDF (3.07 MB)
(June 2004)

**Dating of
Young
Groundwater**
PDF (5.29 MB)
(November
2003)

Tritium/Helium-3



- Half-life 12.3 years; decays to ^3He .
- Atmospheric thermonuclear weapons testing from the 1950's, and especially in the period 1962-1963
- Initial ^3H measured
- Terrigenous He sources
- Dispersion around bomb peak
- Confinement of ^3He



Dating with $^3\text{H}/^3\text{He}$

Measured

$$^3\text{H}_m$$

$$^4\text{He}_m = ^4\text{He}_{eq} + ^4\text{He}_{atm} + ^4\text{He}_{ter}$$

$$\text{Ne}_m = \text{Ne}_{eq} + \text{Ne}_{atm}$$

$$d^3\text{He} = \left(\frac{R}{R_a} - 1 \right) \cdot 100$$

$$^3\text{He}_m = ^3\text{He}_{eq} + ^3\text{He}_{atm} + ^3\text{He}_{ter} + ^3\text{He}_{tri}$$

Some Definitions

$$N_{\text{atm}} = (\text{He/Ne})_{\text{atm}} = 0.288 \text{ for air excess}$$

$$R_{\text{atm}} = ({}^3\text{He}/{}^4\text{He})_{\text{atm}} = 1.384 \times 10^{-6} \text{ for air excess } (R_a)$$

$$R_{\text{ter}} = ({}^3\text{He}/{}^4\text{He})_{\text{ter}} = 2 \times 10^{-8} \text{ for radiogenic helium } (R_{\text{rad}})$$

$$= 1 \times 10^{-5} \text{ for mantle helium } (R_{\text{man}})$$

$$Ne_m = Ne_{eq} + Ne_{atm}$$

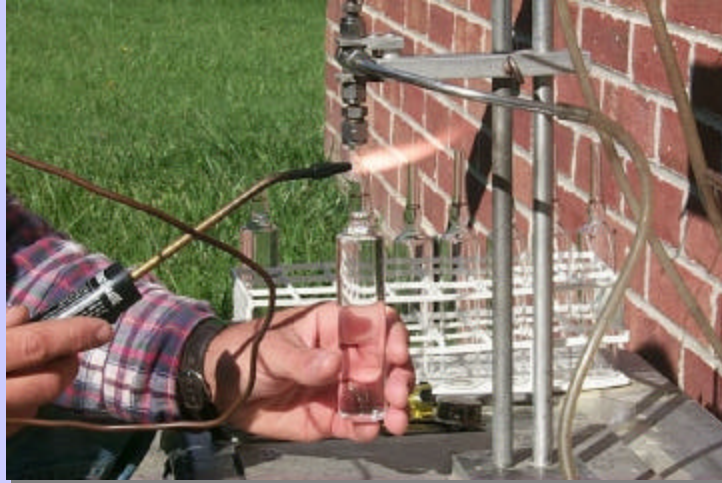
$${}^4\text{He}_m = {}^4\text{He}_{eq} + N_{atm} \cdot Ne_{atm} + {}^4\text{He}_{ter}$$

$${}^3\text{He}_m = R_{eq} \cdot {}^4\text{He}_{eq} + R_{atm} \cdot N_{atm} \cdot Ne_{atm} + R_{ter} \cdot {}^4\text{He}_{ter} + {}^3\text{He}_{tri}$$

$${}^3\text{He}_{tri} = {}^4\text{He}_m \cdot (R_m - R_{ter}) - {}^4\text{He}_{eq} \cdot (R_{eq} - R_{ter}) - N_{atm} \cdot (Ne_m - Ne_{eq}) \cdot (R_{atm} - R_{ter})$$

$$t = \frac{1}{\lambda} \cdot \ln \left(1 + \frac{{}^3\text{He}_{tri}}{{}^3\text{He}_m} \right)$$

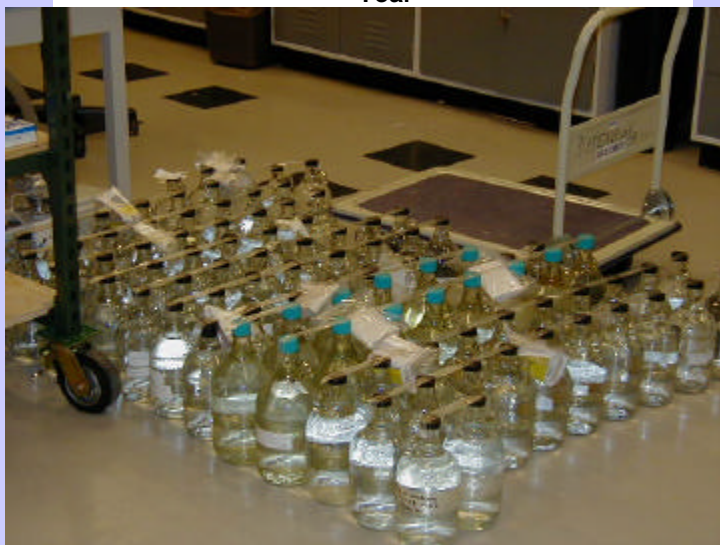
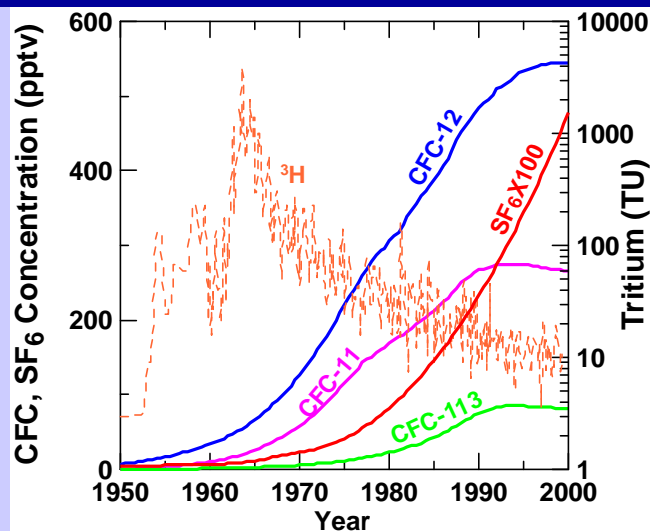
Chlorofluorocarbons



New bottle method of collection.
See <http://water.usgs.gov/lab/cfc>

- CFC-12 (CF_2Cl_2), 1930
- CFC-11 (CFCl_3), 1936
- CFC-113 ($\text{C}_2\text{F}_3\text{Cl}_3$), 1944
- Input smooth, increasing until 1990s (dating range ~1950 to early 1990s).
- Stable in aerobic environments.
- In modern cases: dual ages.
- Use of ratios.
- Can detect post 1940's water.
- Collection/analysis not overly labor intensive.
- Problems with contamination, degradation (anoxic), sorption?

Sulfur Hexafluoride



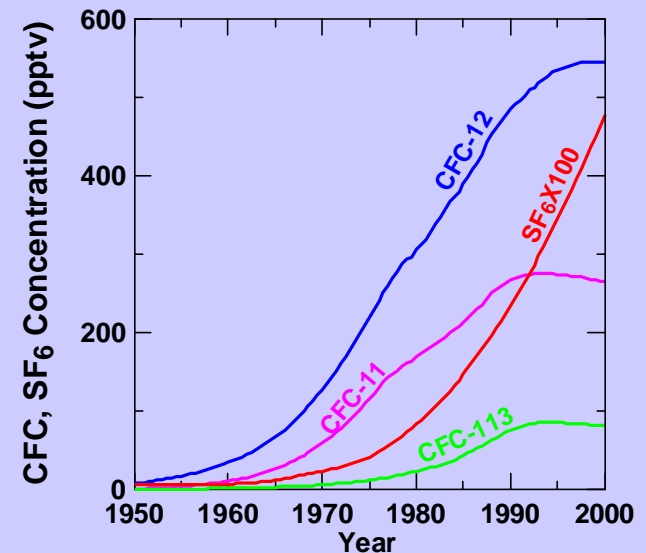
- Electrical insulator in high voltage switches
- First produced 1953
- Very low solubility in water.
- Does not degrade
- Terrigenous source
- Dating range: 1970-modern.
- Smooth input, increasing in air at 6%/yr; 5 pptv today.

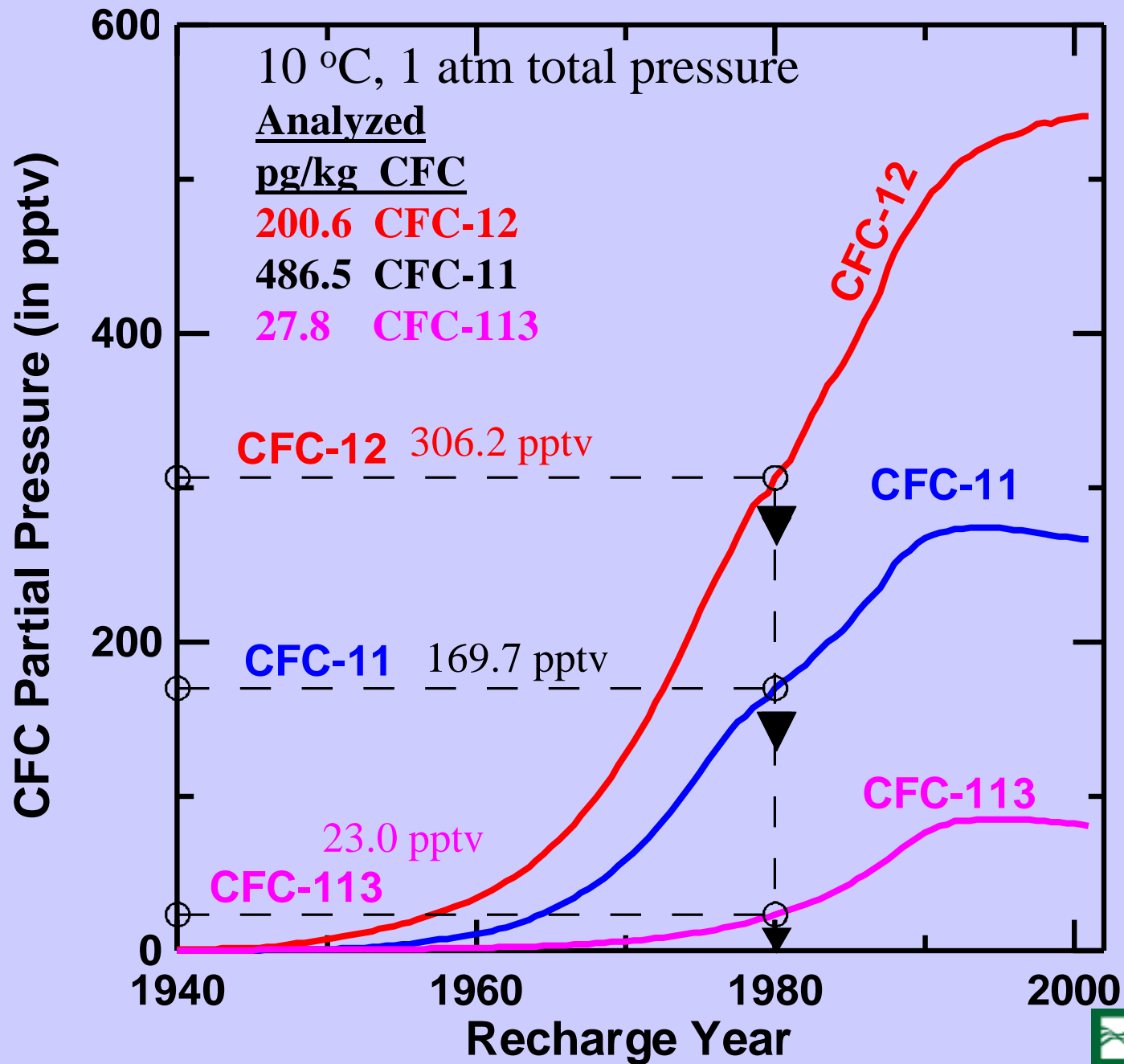
Dating with CFCs and SF₆

- Henry's law solubility $C_i = K_H \times p_i$

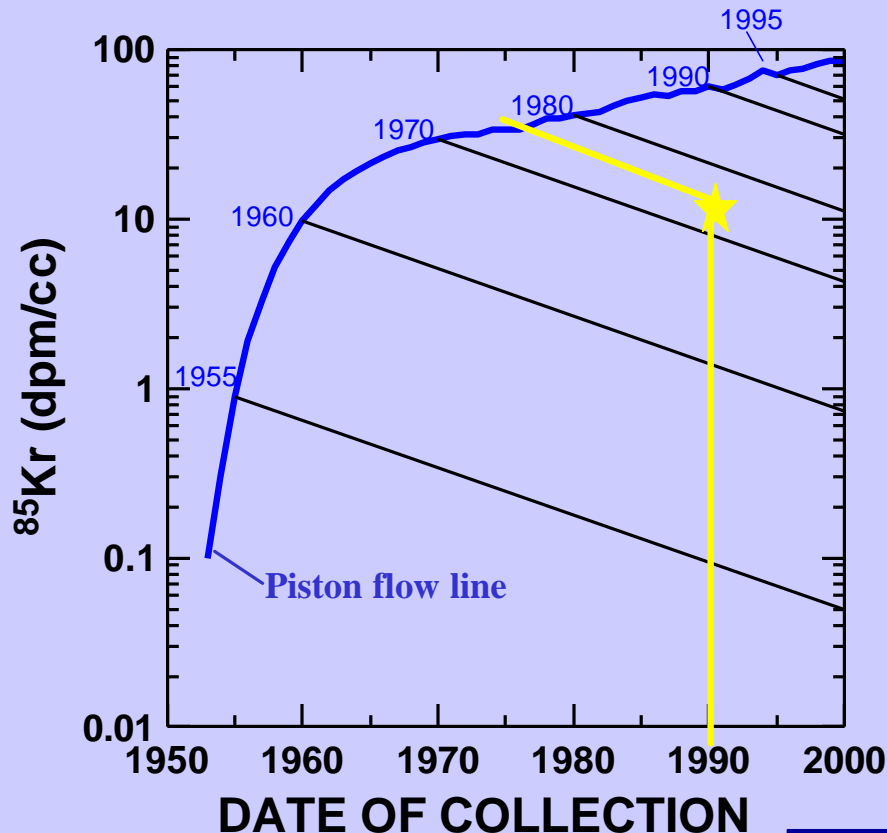
- Requirements:

- Gas-water equilibrium at recharge
- Recharge temperature
- Barometric pressure at recharge
- Knowledge of atmospheric history of the gas





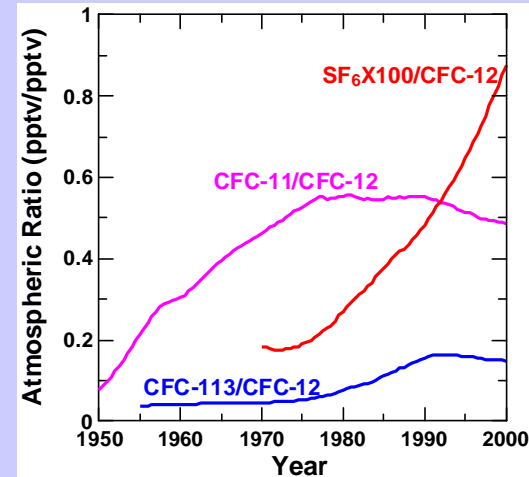
Krypton-85



- Radioactive noble gas with a half-life of 10.76 years.
- ^{85}Kr in the atmosphere has steadily increased since the mid-1950s.
- Interpreted age insensitive to recharge temperature, altitude, etc.
- Does not degrade.
- Atmosphere is only significant source.
- Although difficult to collect and analyze, even a few ^{85}Kr can often help resolve questions.

Example: sample collected in 1990 had a Specific Activity of Kr-85 of 10 dpm/cc. This would have a Piston flow recharge date of about 1975.

Tracer method of dating the young fraction in binary mixtures of young and old



- CFC pptv ratio defines age of young fraction.
 - % young water $(\text{pptv}_{\text{(measured)}} / \text{pptv}_{\text{(ratio year)}}) \times 100$
-
- Cannot date outside range for ratio.
 - Cannot use if one of the CFCs in the selected ratio is “contaminated”, even if ratio is “in range”.
 - Ratio-based age must be less than (younger than) apparent (piston flow) model ages for both CFCs in the ratio.

$^3\text{H}/^3\text{He}$ Age Applies to that of the Young Fraction in Simple Binary Mixtures

$^3\text{H}/^3\text{He}$ Age is based on an isotope ratio:
($^3\text{H}^0/^3\text{H}_m$)

$$t = \frac{1}{\lambda} \cdot \ln \left(1 + \frac{^3\text{He}_{tri}}{^3\text{H}_m} \right)$$

λ = decay constant = $\ln 2/t_{1/2} = 0.05635 \text{ year}^{-1}$

$t_{1/2} = 12.3 \text{ years}$

Limitations

$^3\text{H}/^3\text{He}$

- Cost
- Terrigenous He
- Bubbles, Gas-stripping, confinement
- Mixing

SF_6

- Terrigenous SF_6
- Mixing

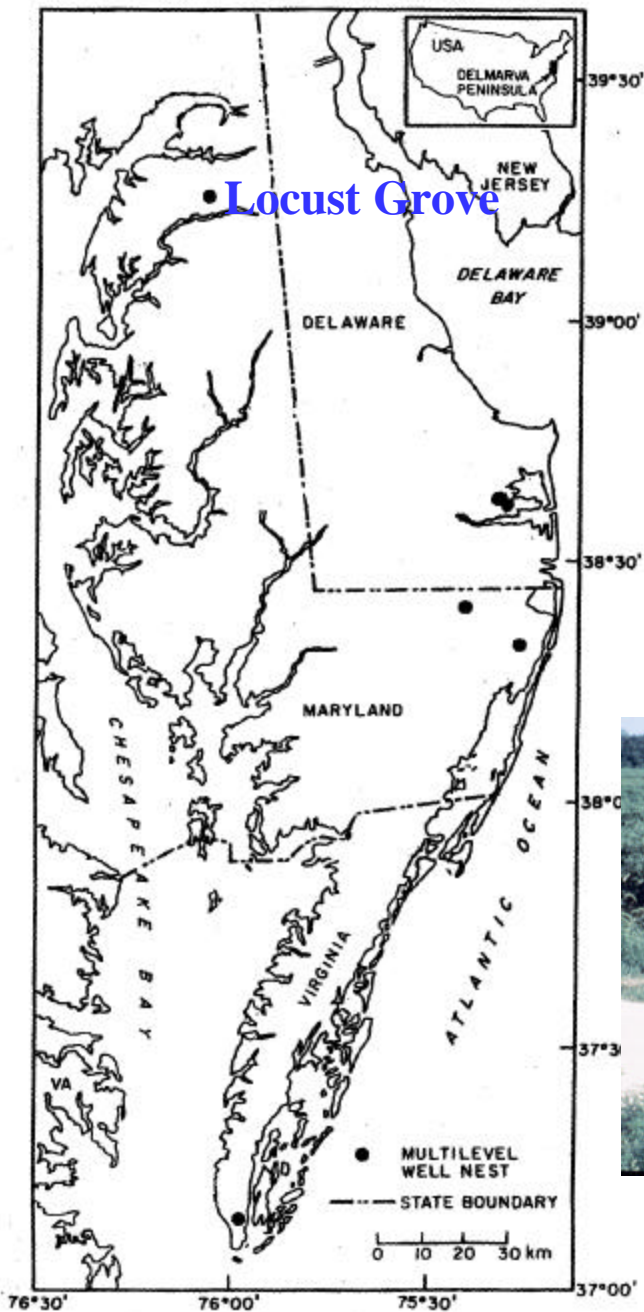
CFCs

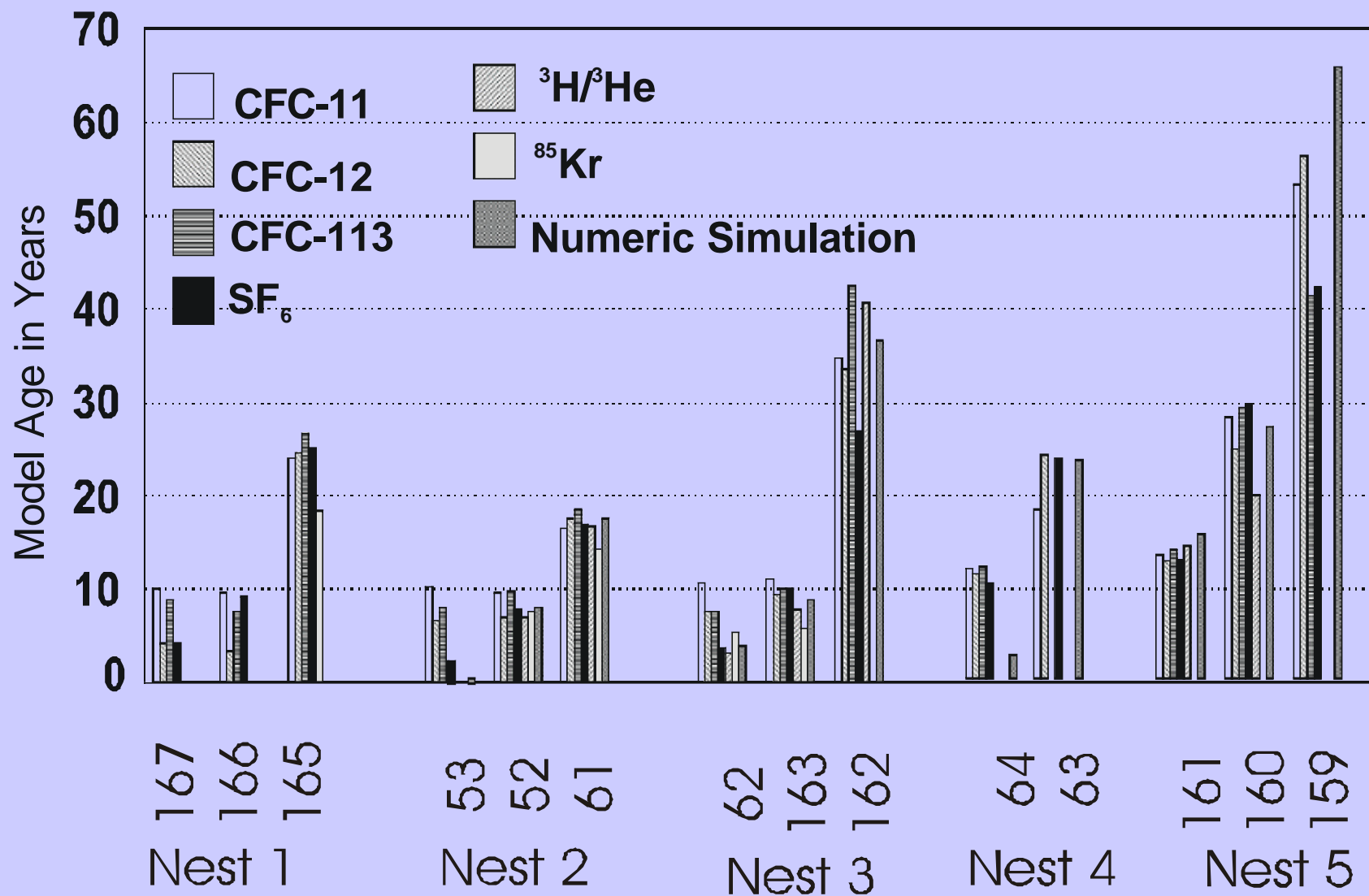
- Contamination
- Degradation
- Mixing

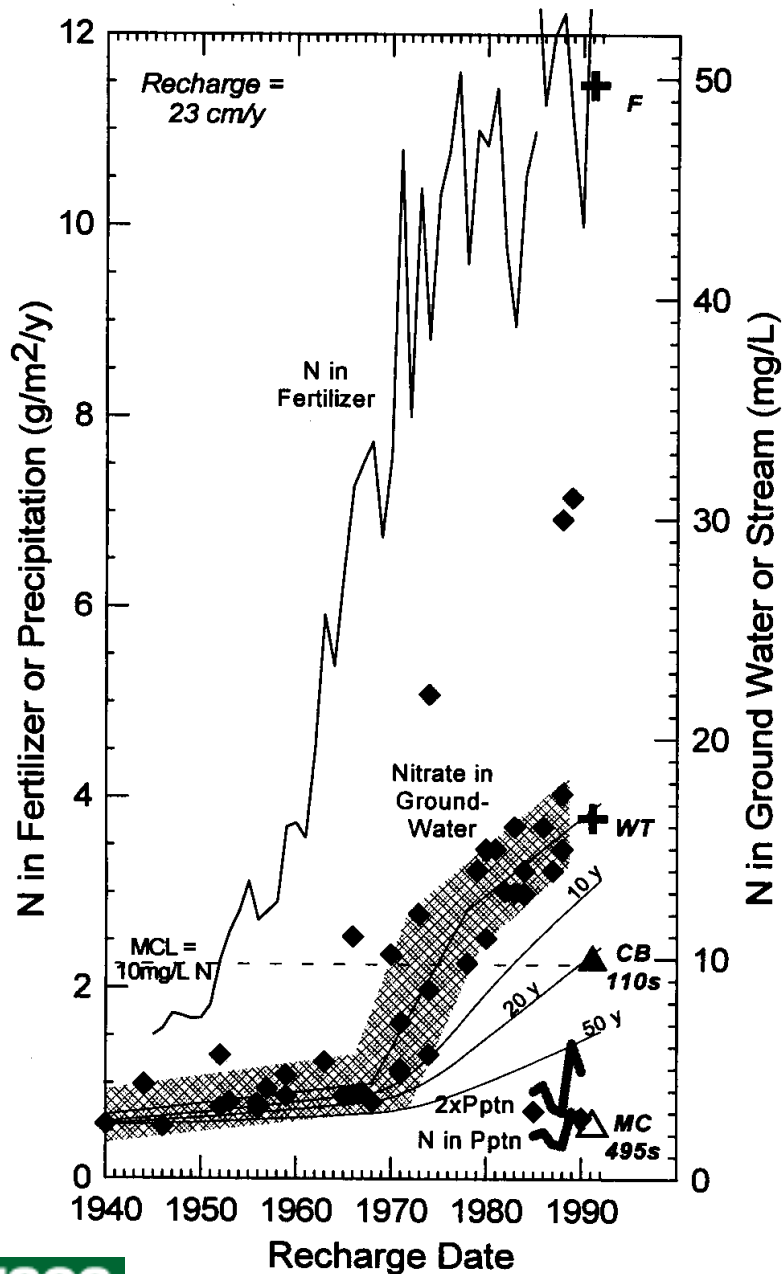
^{85}Kr

- Very difficult collection and analysis
- No labs available to us

Delmarva Peninsula Atlantic Coastal Plain







- CFC ages and NO_3 : 40-yr record of NO_3 recharge rate.
- Records increase in fertilizer application from the 1970's.
- 20-35 % of applied fertilizer reached the aquifer.
- Mean residence time of 20 yrs for gw discharge to local streams.

Böhlke and Denver (1995)

Blue Ridge Mountains of Virginia



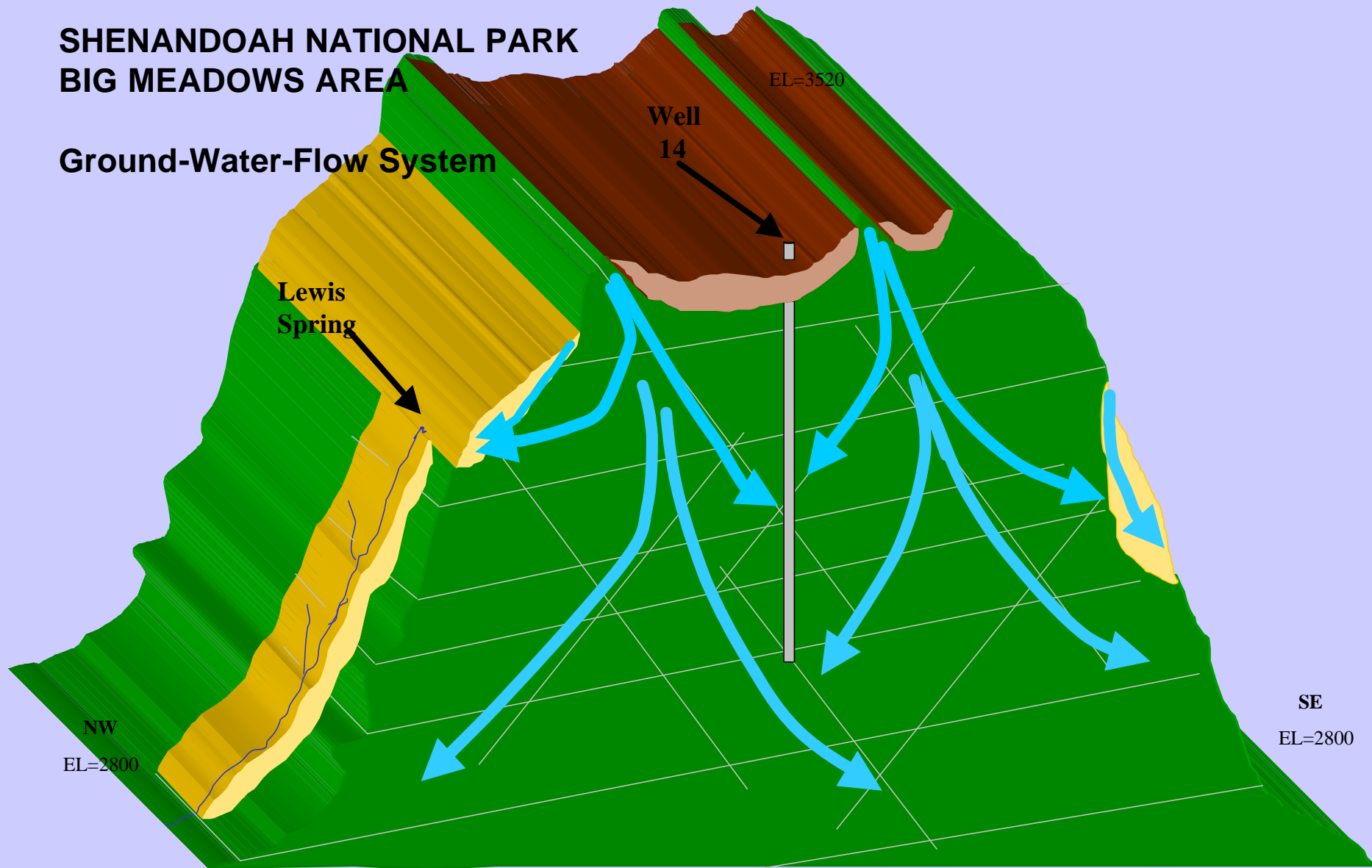
Plummer, L.N., Busenberg, E., Böhlke, J.K., Nelms, D.L., Michel, R.L., and Schlosser, P., 2001, Groundwater residence times in Shenandoah National Park, Blue Ridge Mountains, Virginia, USA: A multi-tracer approach. *Chemical Geology*, v. 179/1-4, p. 93-111.

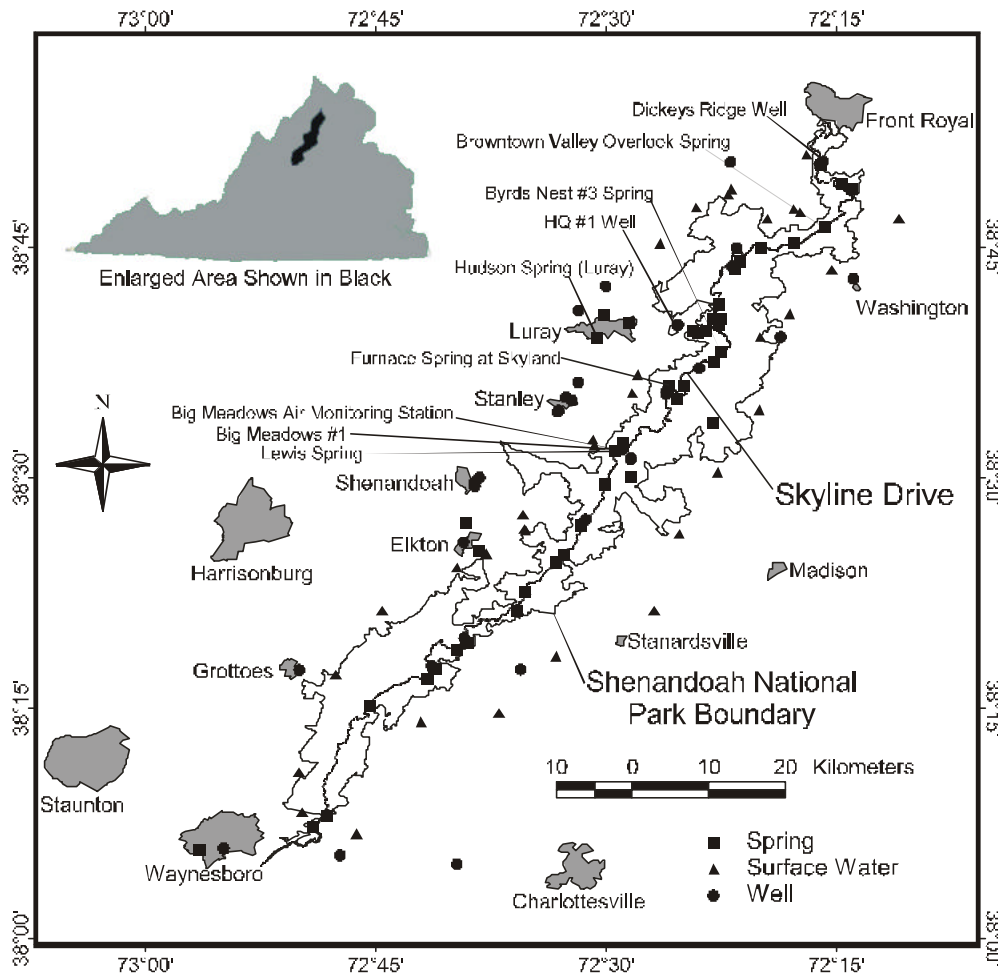
Background

- 800 km²; elev. 170-1230 m
- Annual precipitation averaged 114 cm; mean annual temperature, 7.8 °C in high altitude, central part.
- Precambrian to Cambrian fractured crystalline rocks, metabasalts with thin cover of colluvium and residuum.
- Two largest springs have maximum discharges of 20 and 11 l/sec. Most with max. discharge of < 2 l/sec.
- In drought, sprs. about 10% of max.
- wells produce < 6 l/sec, and typically < 1 l/sec.
- 34 springs, 15 wells: 1996 (wet season), 1997 (dry season).
- Shallow recharge through residuum and colluvium recharge fracture system which has low storage.
- CFCs, ³H/³He, SF₆, ³⁵S, stable isotopes, dissolved N₂, Ar.

SHENANDOAH NATIONAL PARK BIG MEADOWS AREA

Ground-Water-Flow System



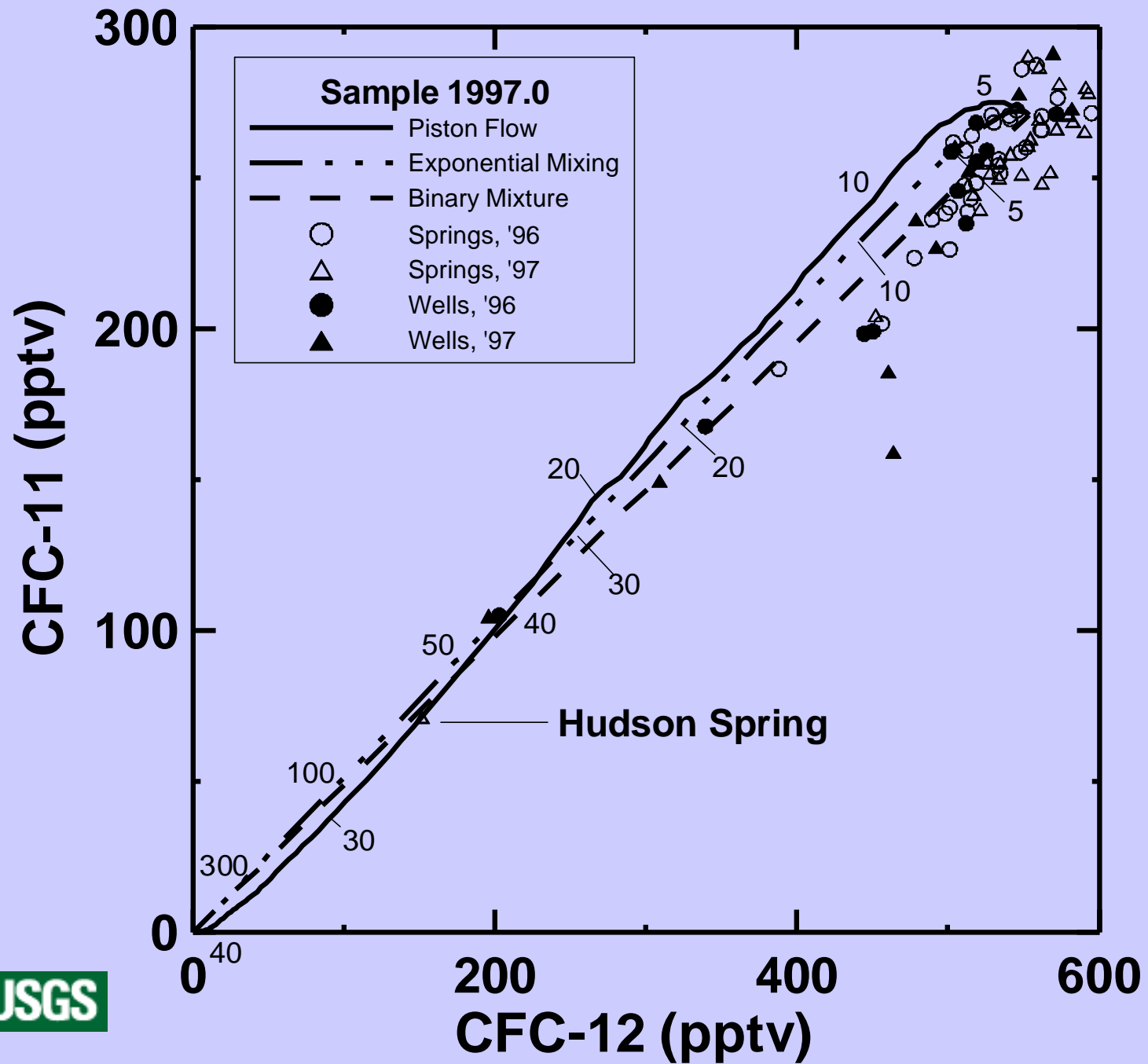


Lewis Mountain Spring

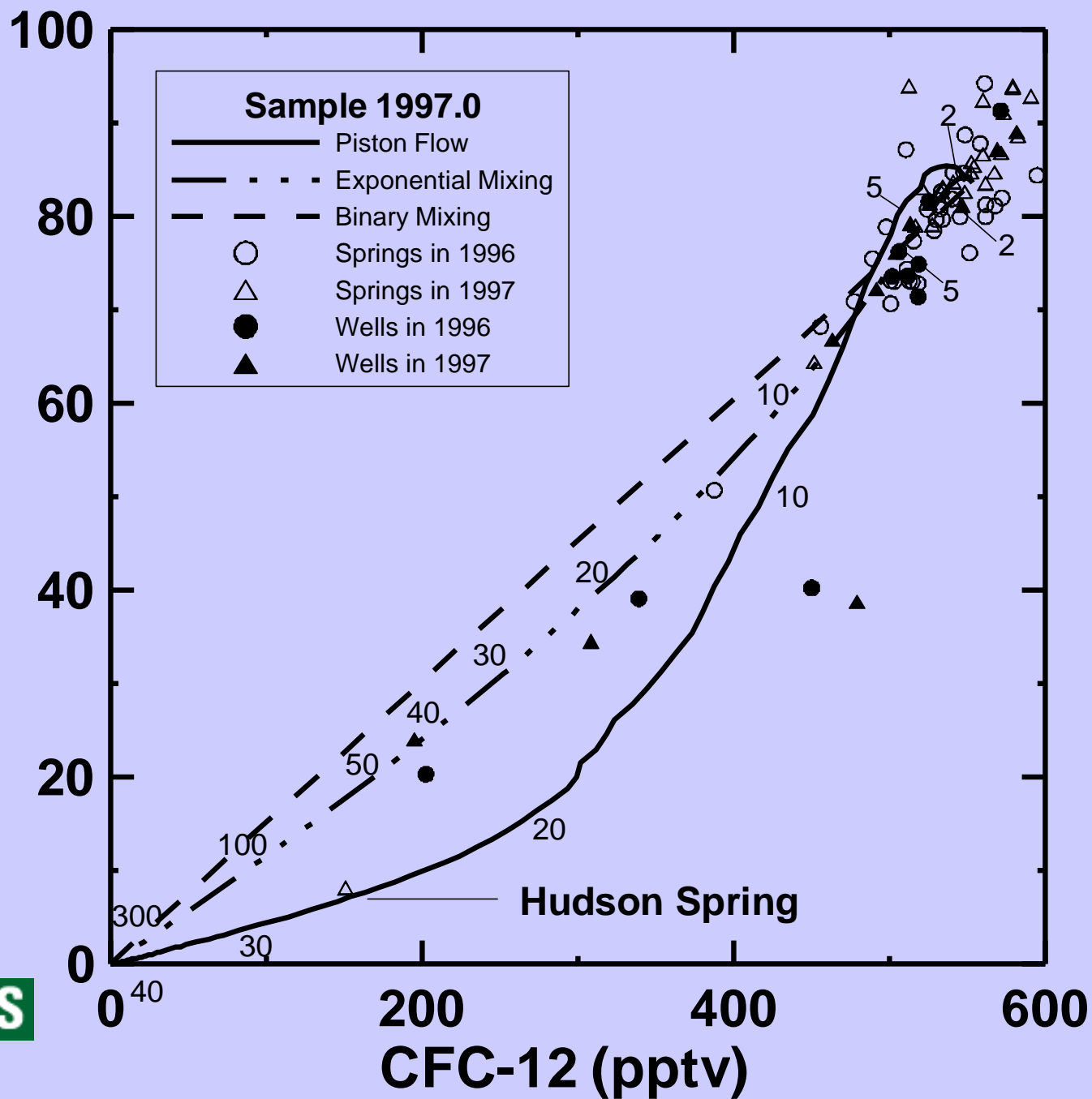


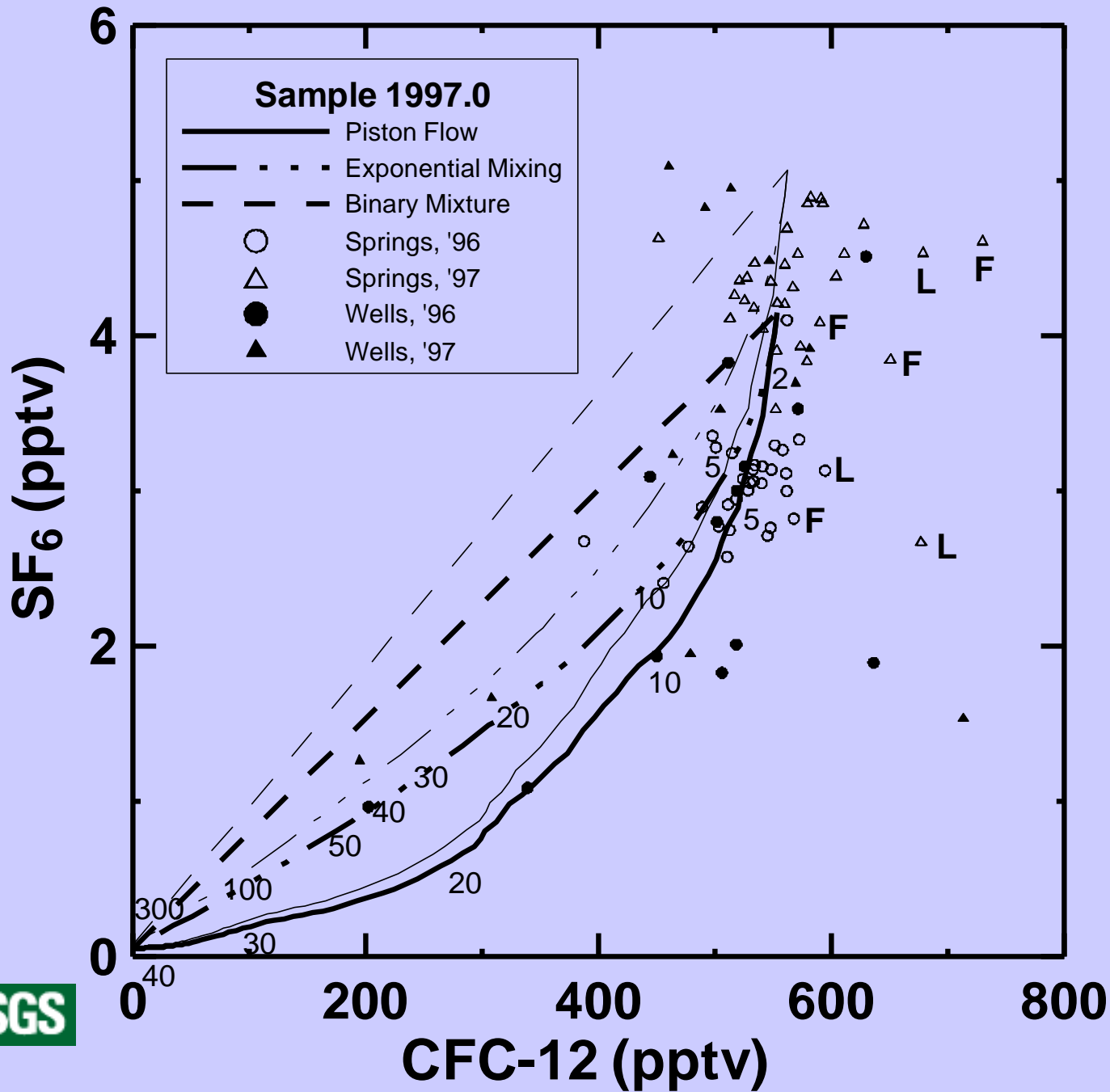
Sampling from a Typical Spring Box and Weir

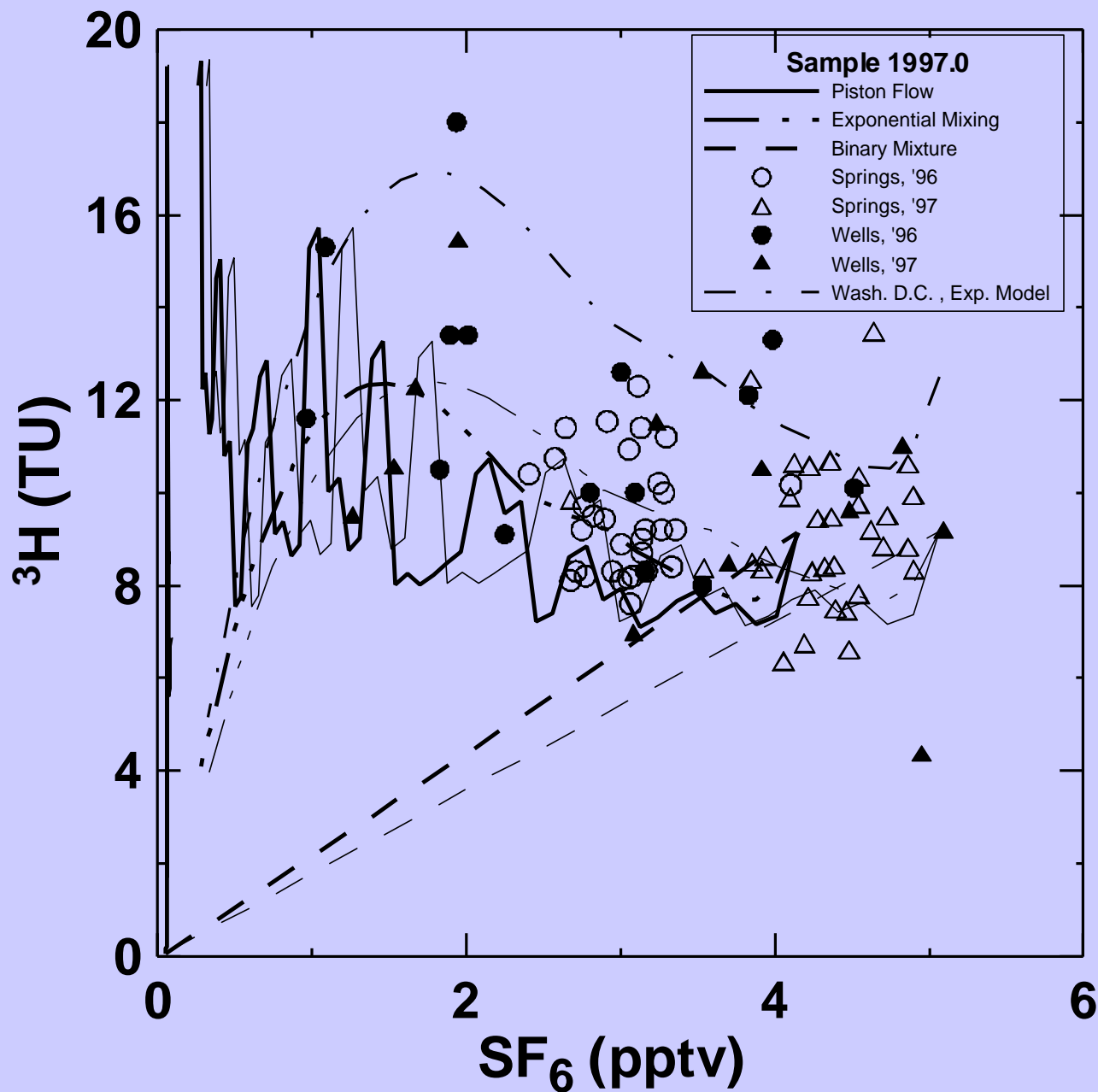




CFC-113 (pptv)



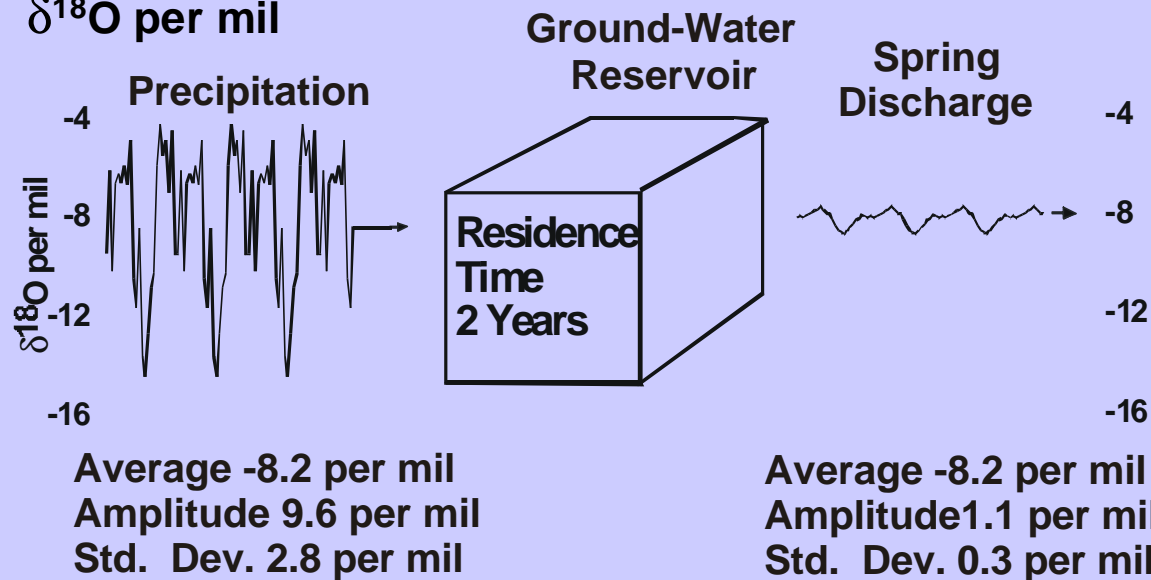
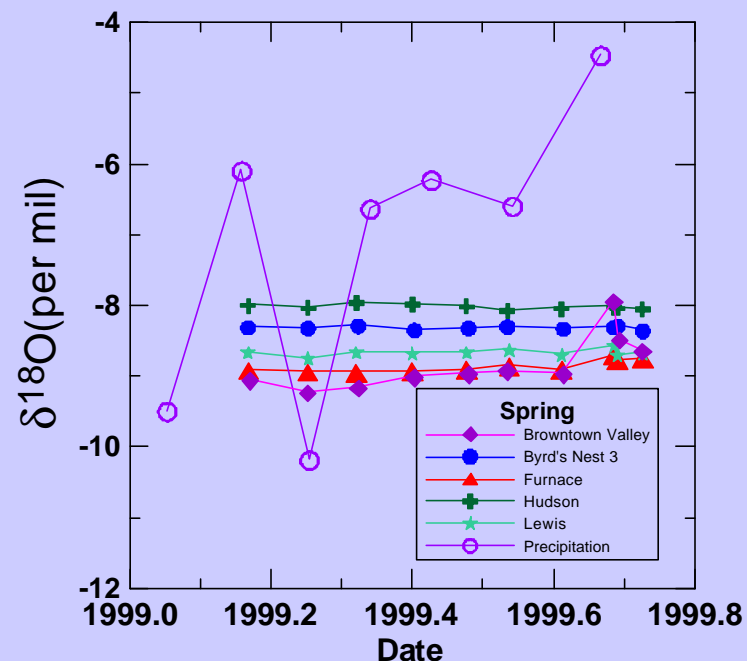
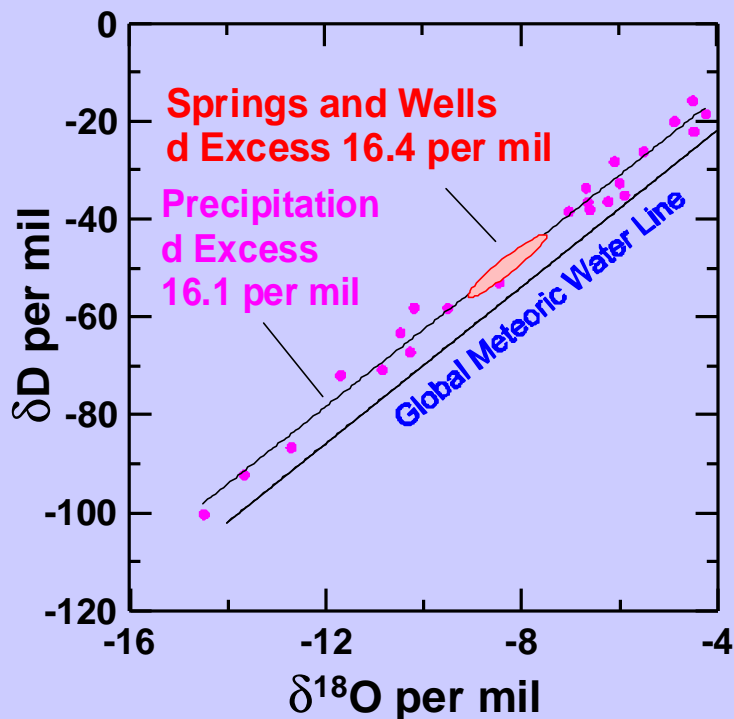




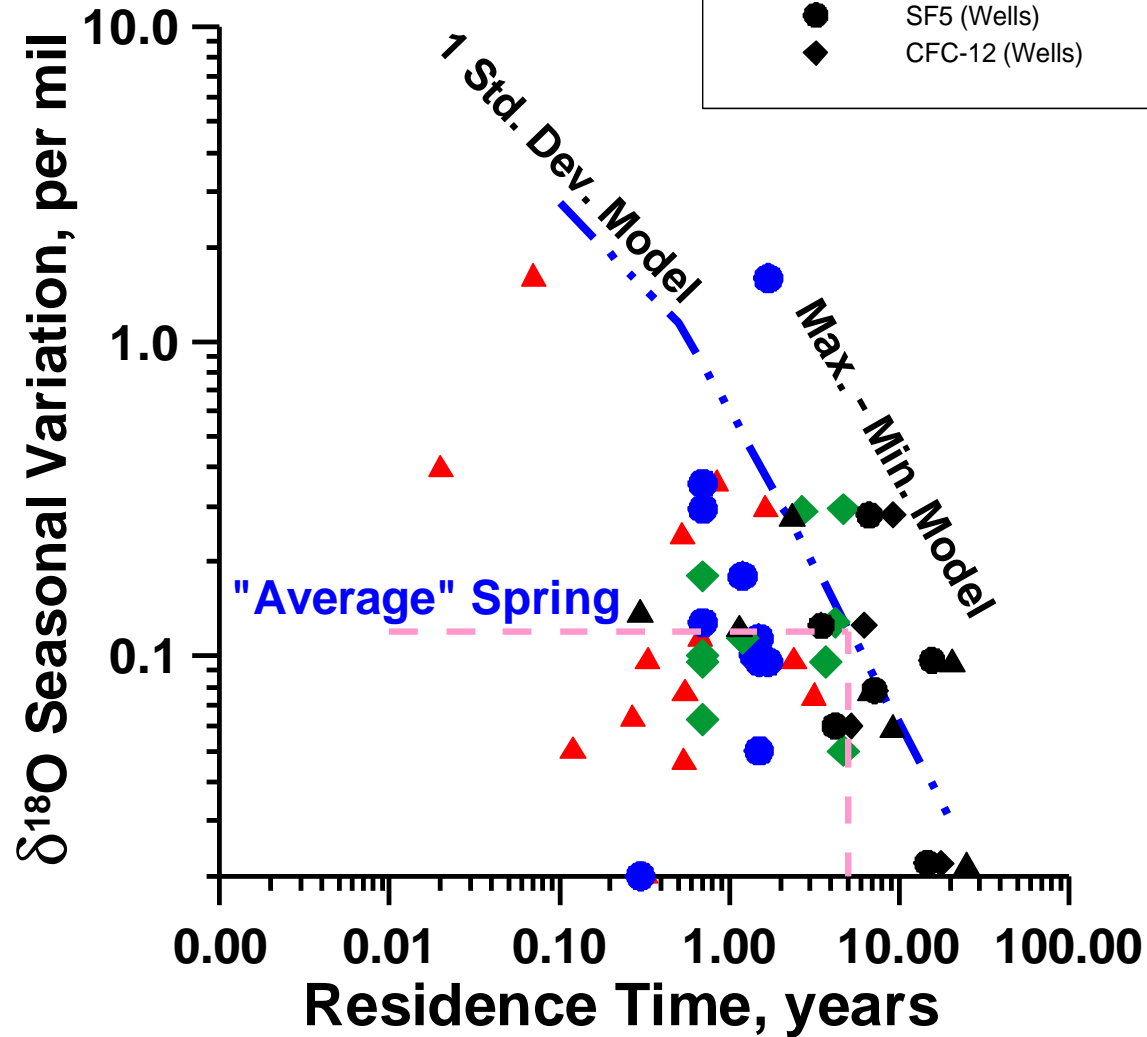
Two ^3H input functions:

- Wash. DC (upper)
- Wash. DC scaled to SNP (lower)

Also 0-2 cc/L of excess air

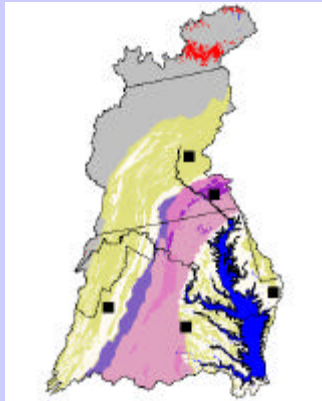


Average Seasonal
 $\Delta \delta^{18}\text{O} = 0.12$ per mil
Average Age ~ 5 years



Conclusions from Blue Ridge (SNP)

- $\delta^{18}\text{O}$ data: Mean residence time of 5 years. (Range 0-10 yrs.)
- Ages based on CFCs and SF_6 young; generally consistent with stable isotope data; do not include uz travel time.
- Ages based on $^3\text{H}/^3\text{He}$ biased young (0-2 yrs, most 0)
- Water from wells have ages of 0 to 25 yrs. $^3\text{H}/^3\text{He}$ dating works well for these (and applies to the young fraction).
- Some wells discharge mixtures; ratios of CFCs can define age of young fraction and percent young water in mixture.
- Excess CFCs-- Anthropogenic sources? Evidence of shallow recharge?



Chesapeake Bay Watershed

Region of 165,000 km² over parts of NY, PA, MD, DE, VA, and WV.

Evaluate residence times and nitrate transport in groundwater discharging to streams in the watershed.

Assess lag time between changes at the land surface and the response in the base-flow component of groundwater discharge to the bay.

Lindsey and others, 2003, USGS WRIR 03-4035,
<http://pa.water.usgs.gov/reports/wrir03-4035.pdf>



The Chesapeake Bay Watershed

64,000 Square Miles of Land, Water, and People

"A Better Bay Through Better Science"

1997

Produced by the USGS from a mosaic of Landsat satellite imagery acquired from 1980-1984



Chesapeake Bay Watershed
 1:500,000 Scale Map



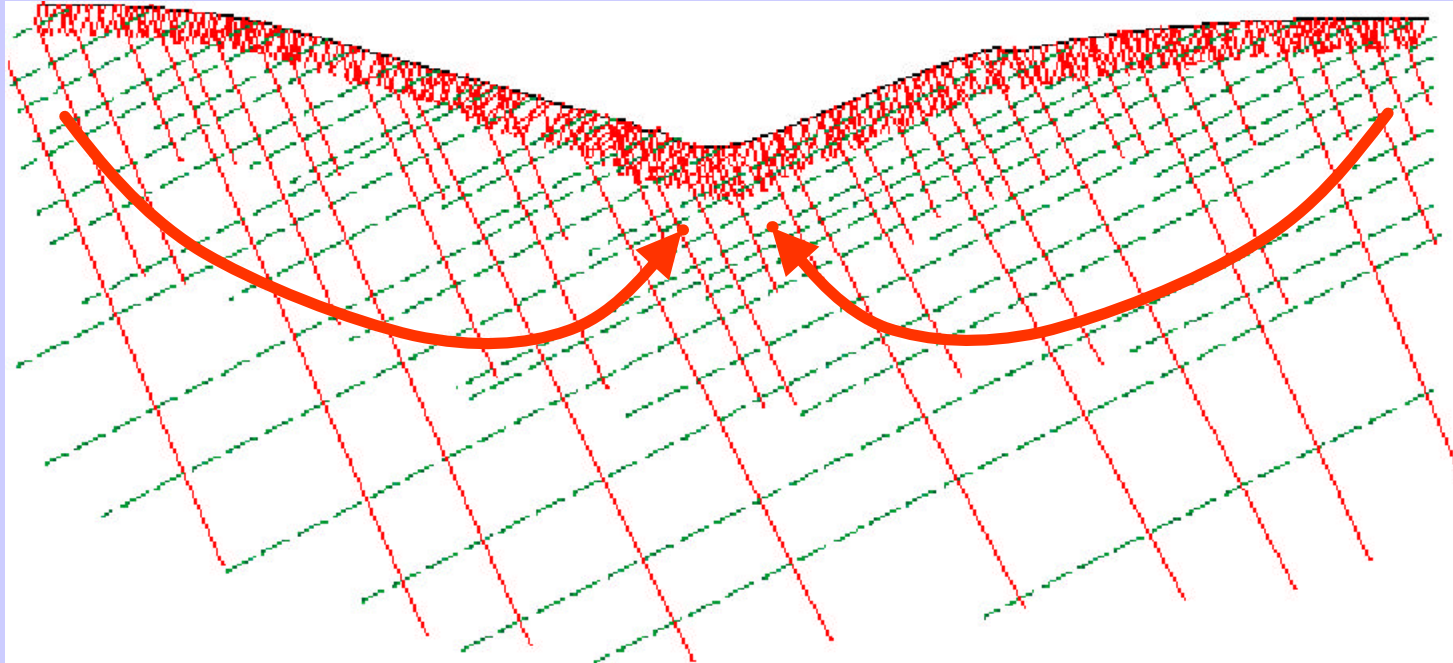
U.S. Geological Survey
<http://www.usgs.gov>



Layered fracture density model (Gburek, Folmar, and Urban (1998))

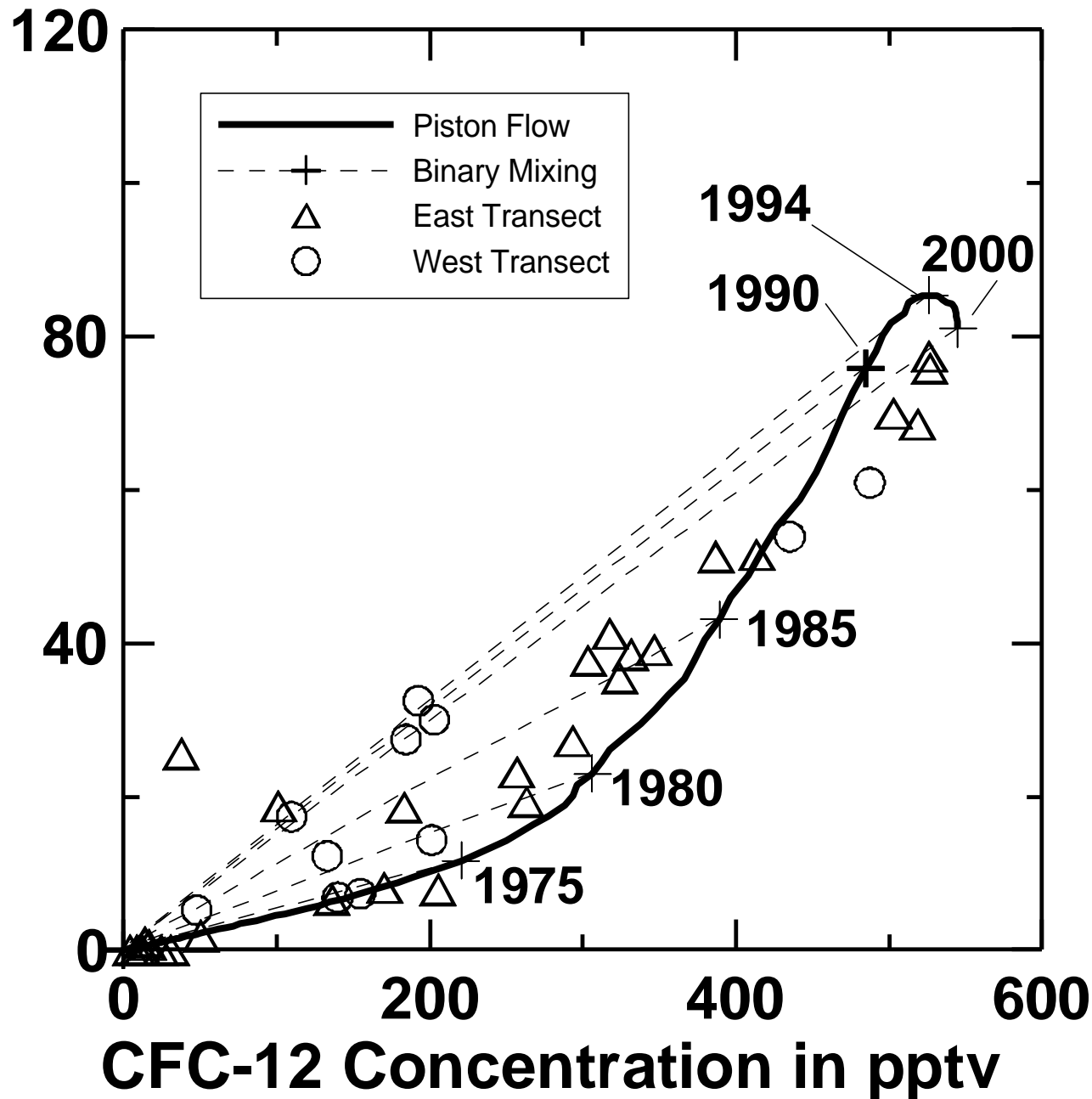
Depth (ft)

0
20
40
60
80



Burton et al. 2002

CFC-113 Concentration in pptv

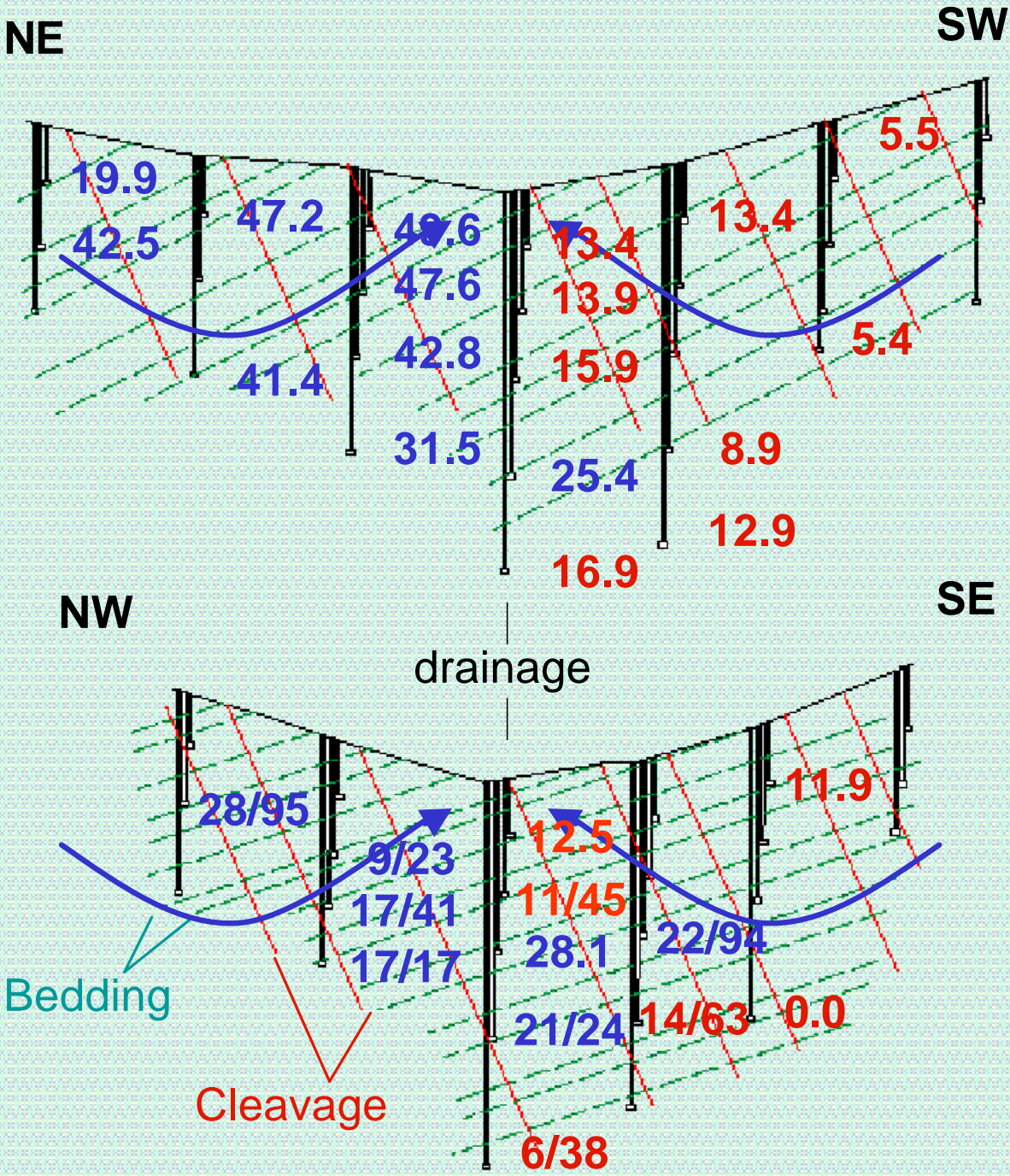


East Piezometer Transect

- Preferred CFC-12 ages NE of stream are older...
- ..than those underneath and SW of stream
- ..although these ages are really mixture ages!

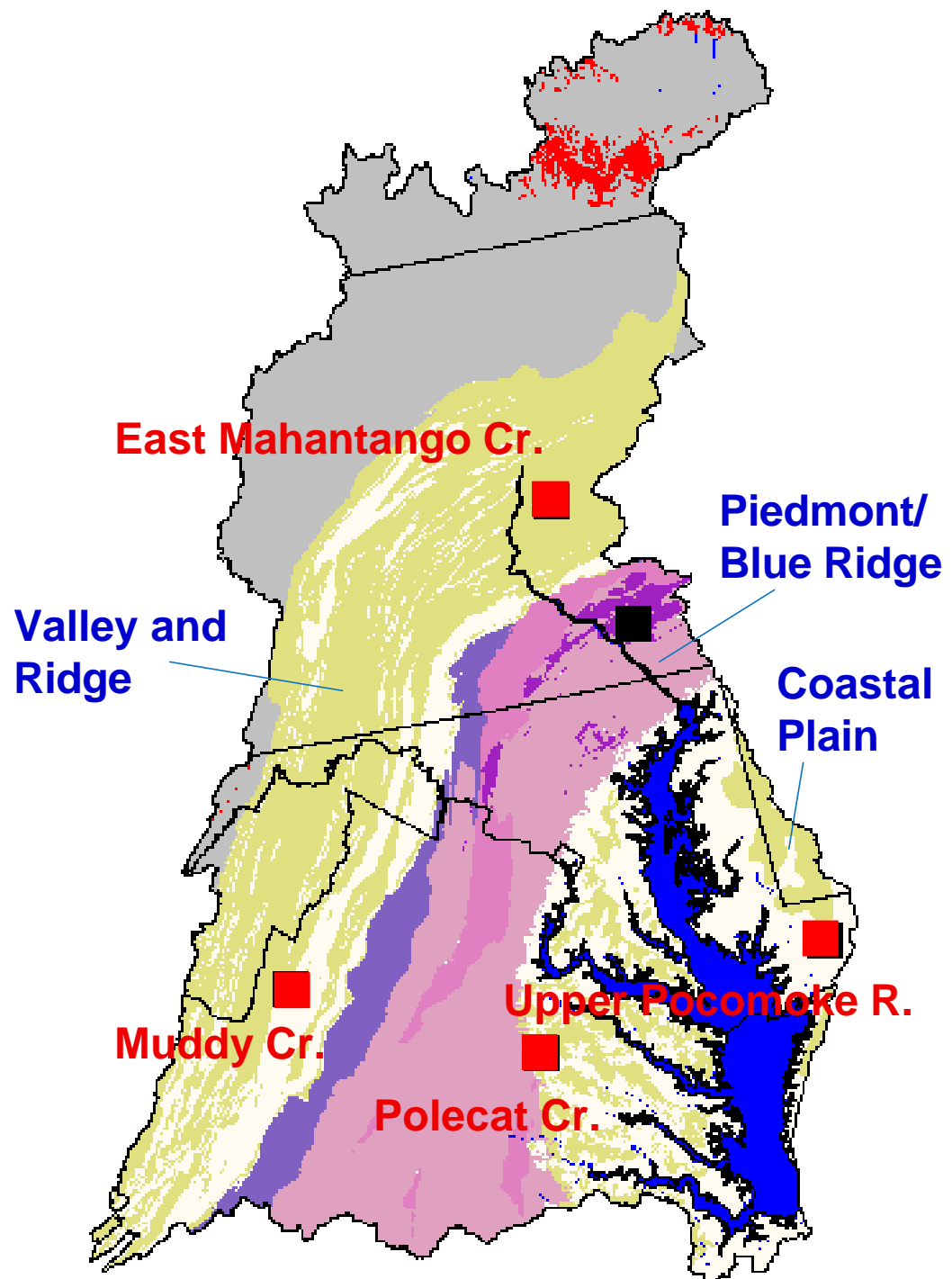
West Piezometer Transect

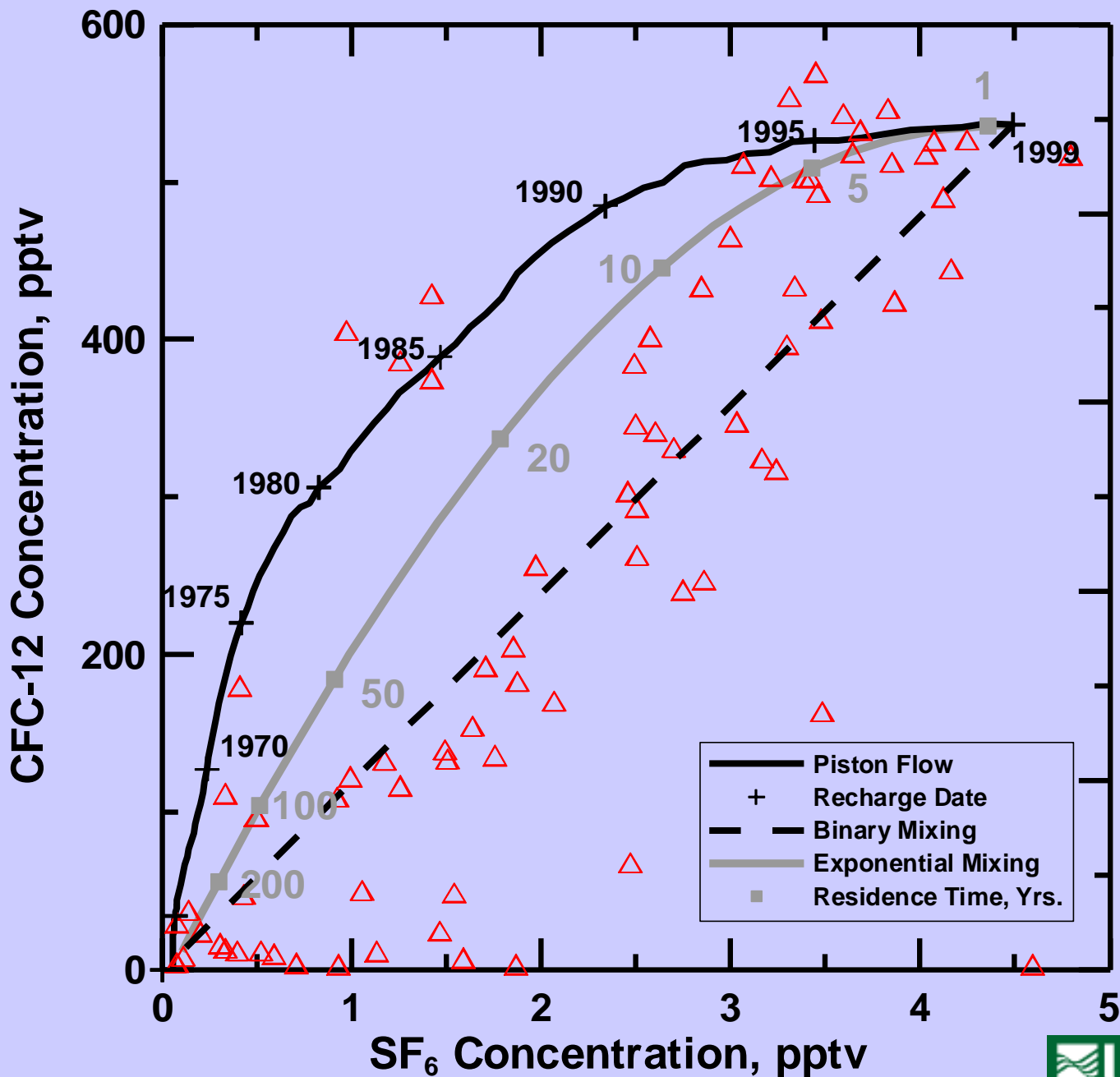
- Ages are even more mixed, but % of young water (2nd #) shows older waters generally prevailing to NW
- ...and younger to SE

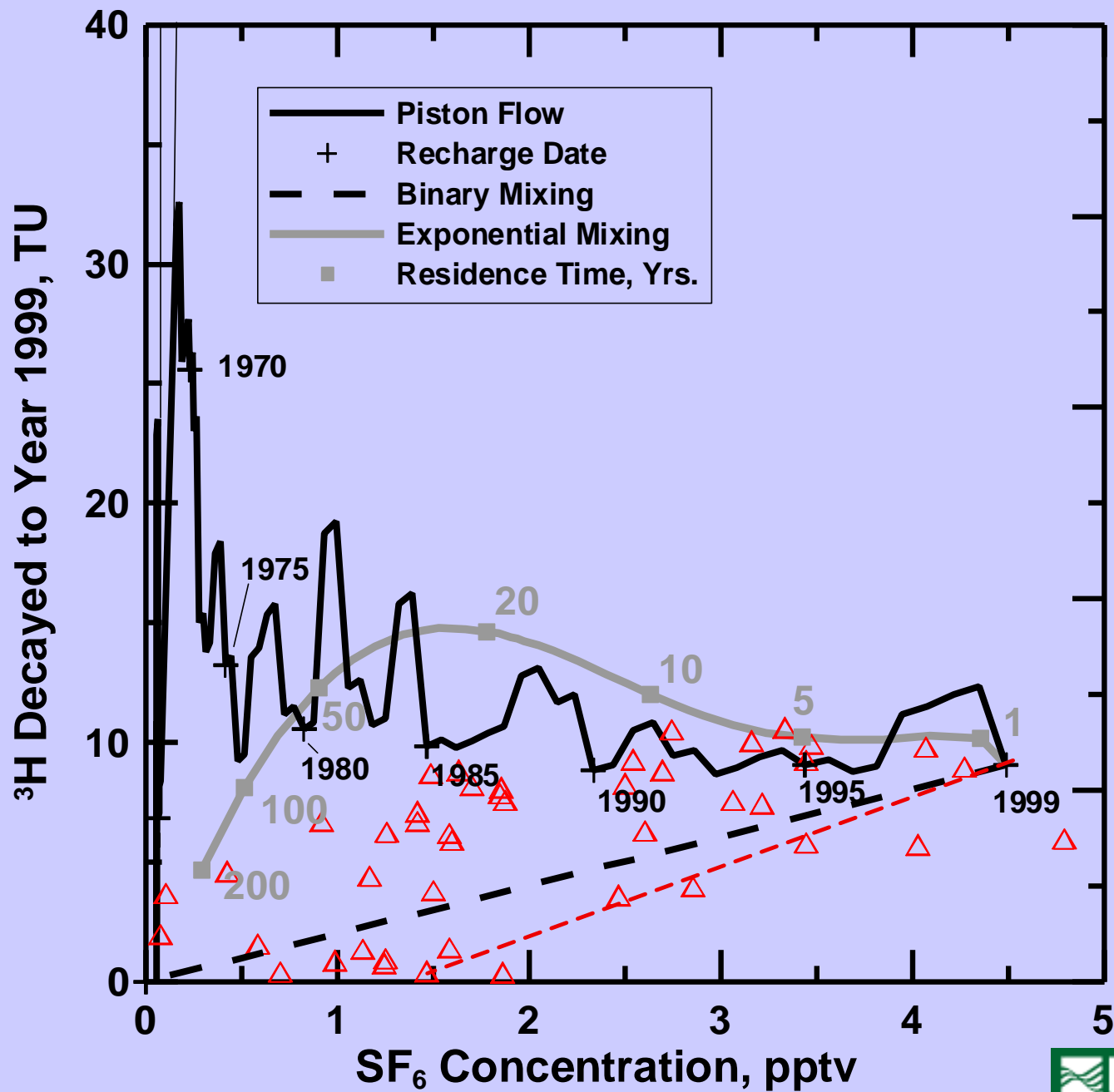


USGS Chesapeake Bay watershed nutrient study

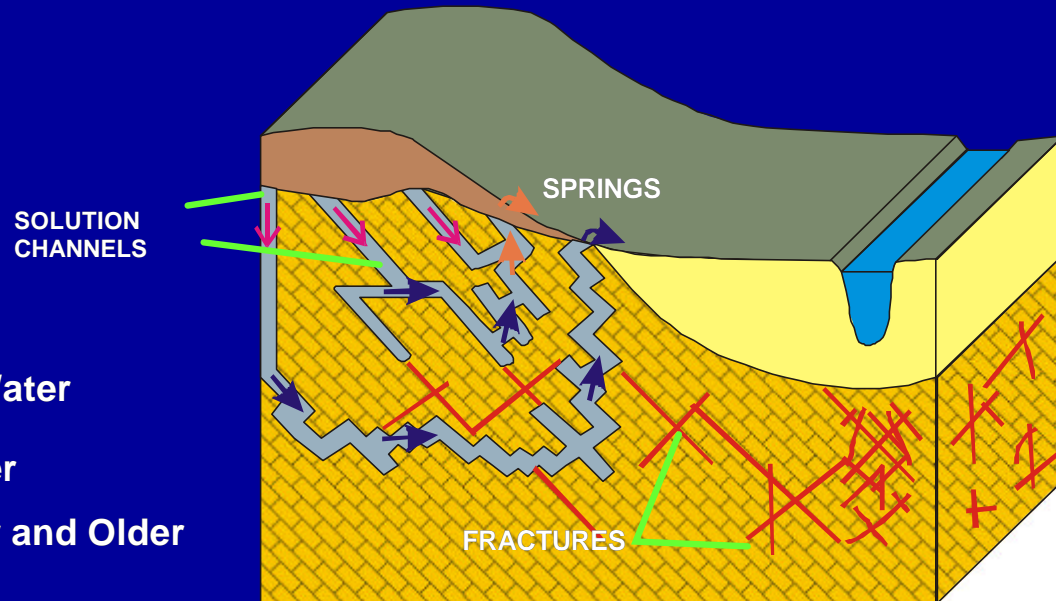
- SF_6 vs ^3H , CFCs
- Monitoring wells, springs from watersheds







Hydrogeology of Carbonate Terranes of the Valley & Ridge Province

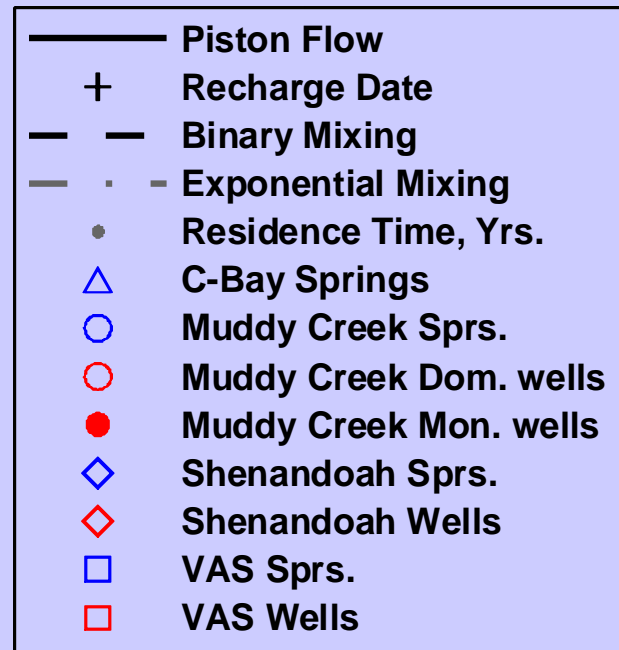
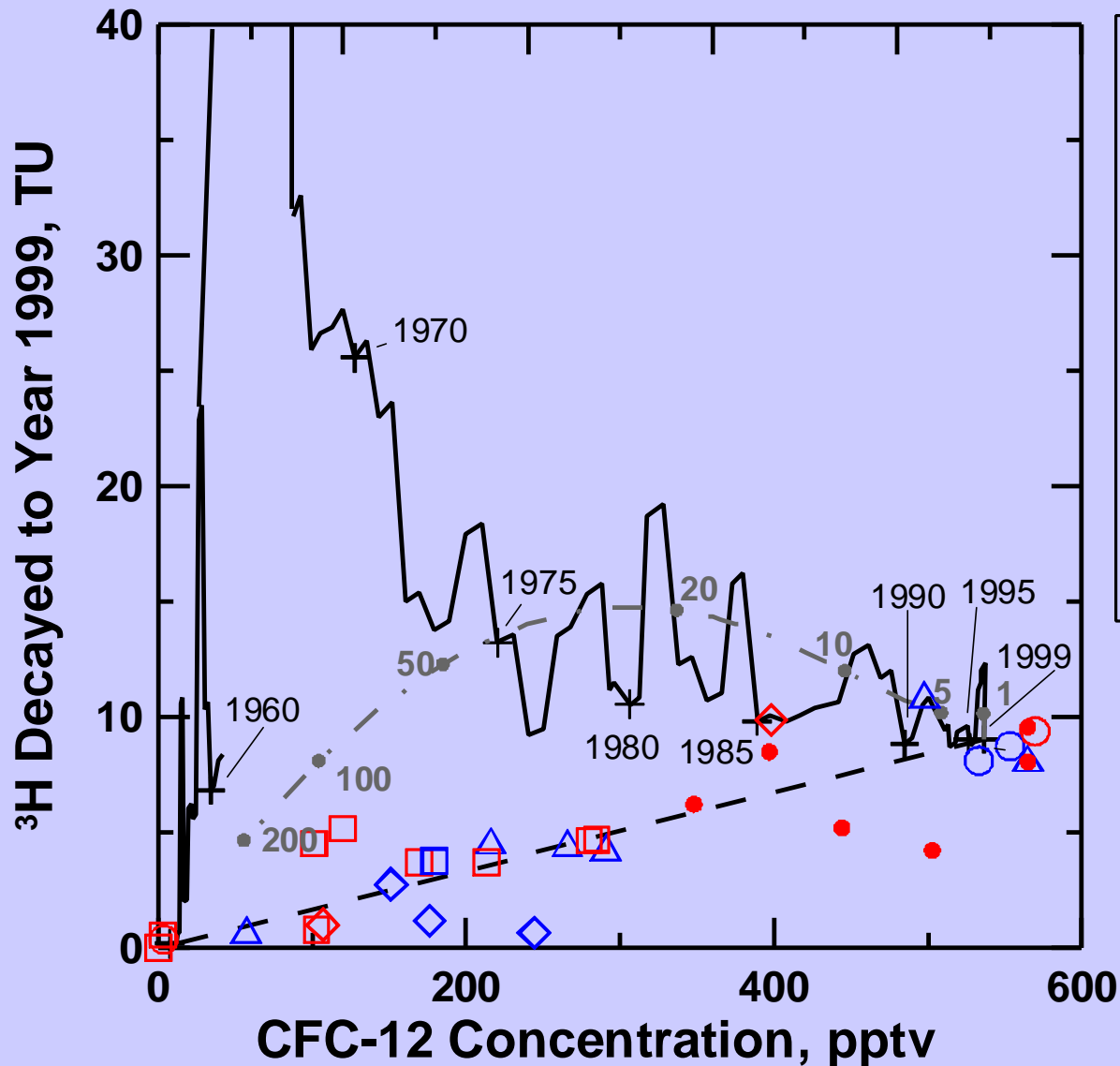


Modified from Brahana and others, 1986

NOT TO SCALE

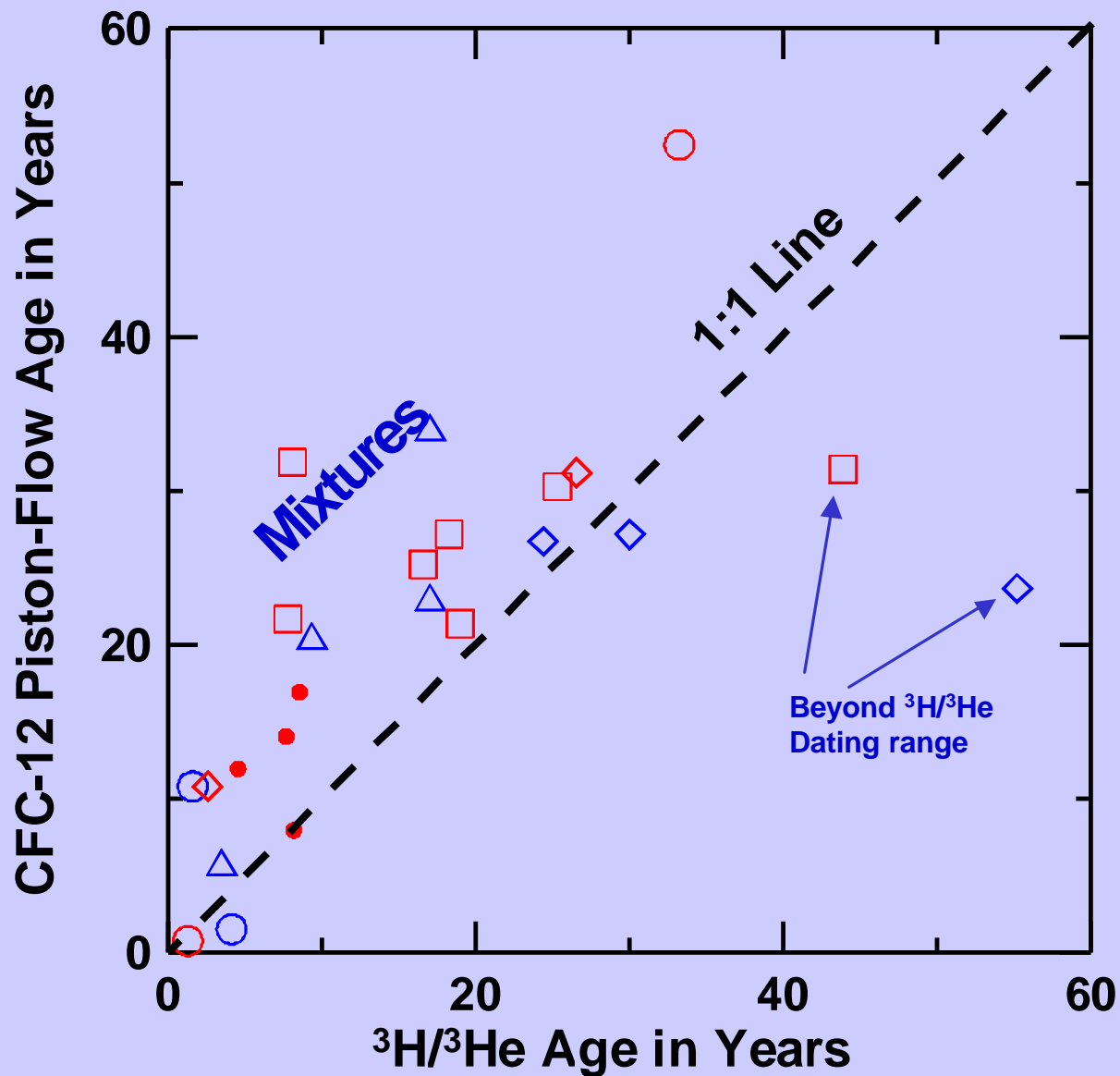


Blue Hole Spr., VA



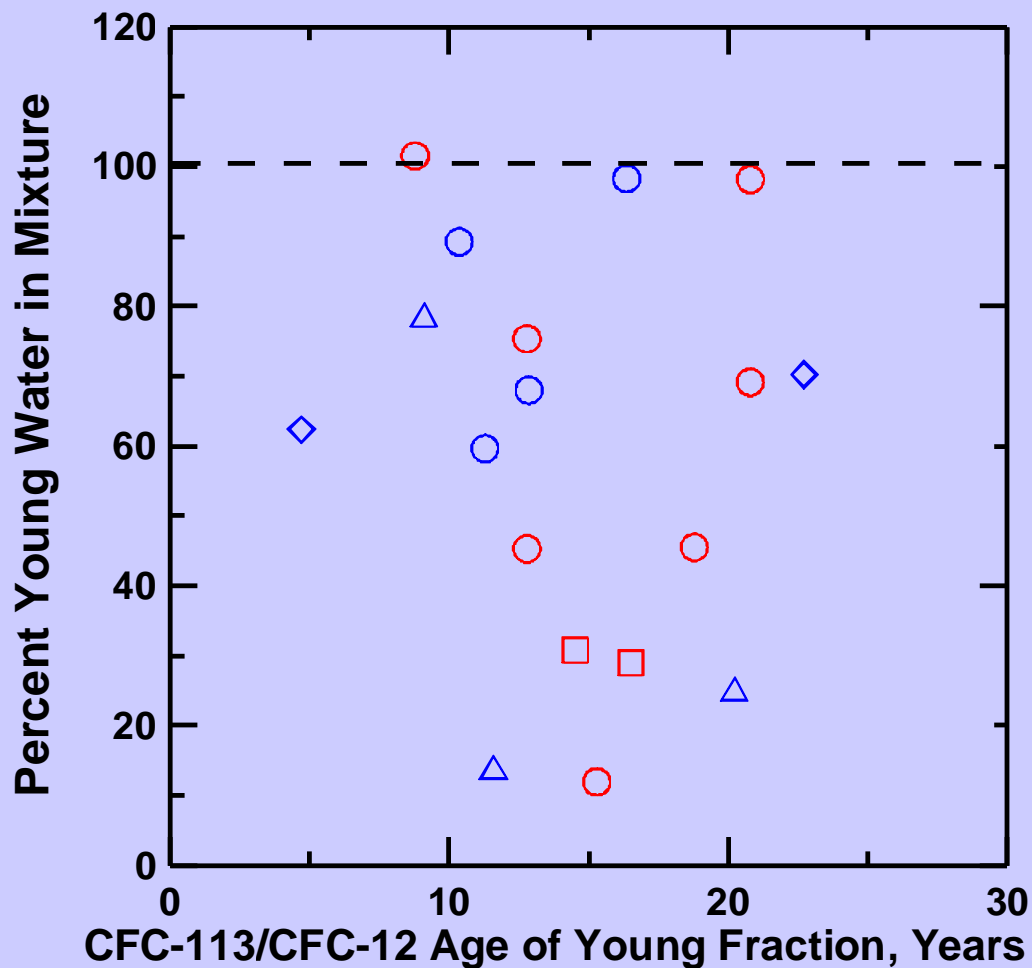
Tritium vs CFC-12

- Young, piston flow
- Binary mixtures of young and old
- Few CFC-12 contamination



- Valley and Ridge Carbonates**
- △ C-Bay Sprs.
 - Muddy Creek Sprs.
 - Muddy Creek Dom. wells
 - Muddy Creek Mon. wells
 - ◇ Shenandoah Sprs.
 - ◇ Shenandoah wells
 - VAS Sprs.
 - VAS wells

- 2 samples beyond dating range of $^3\text{H}/^3\text{He}$.
- 9 samples may be unmixed.
- 13 samples look like mixtures.

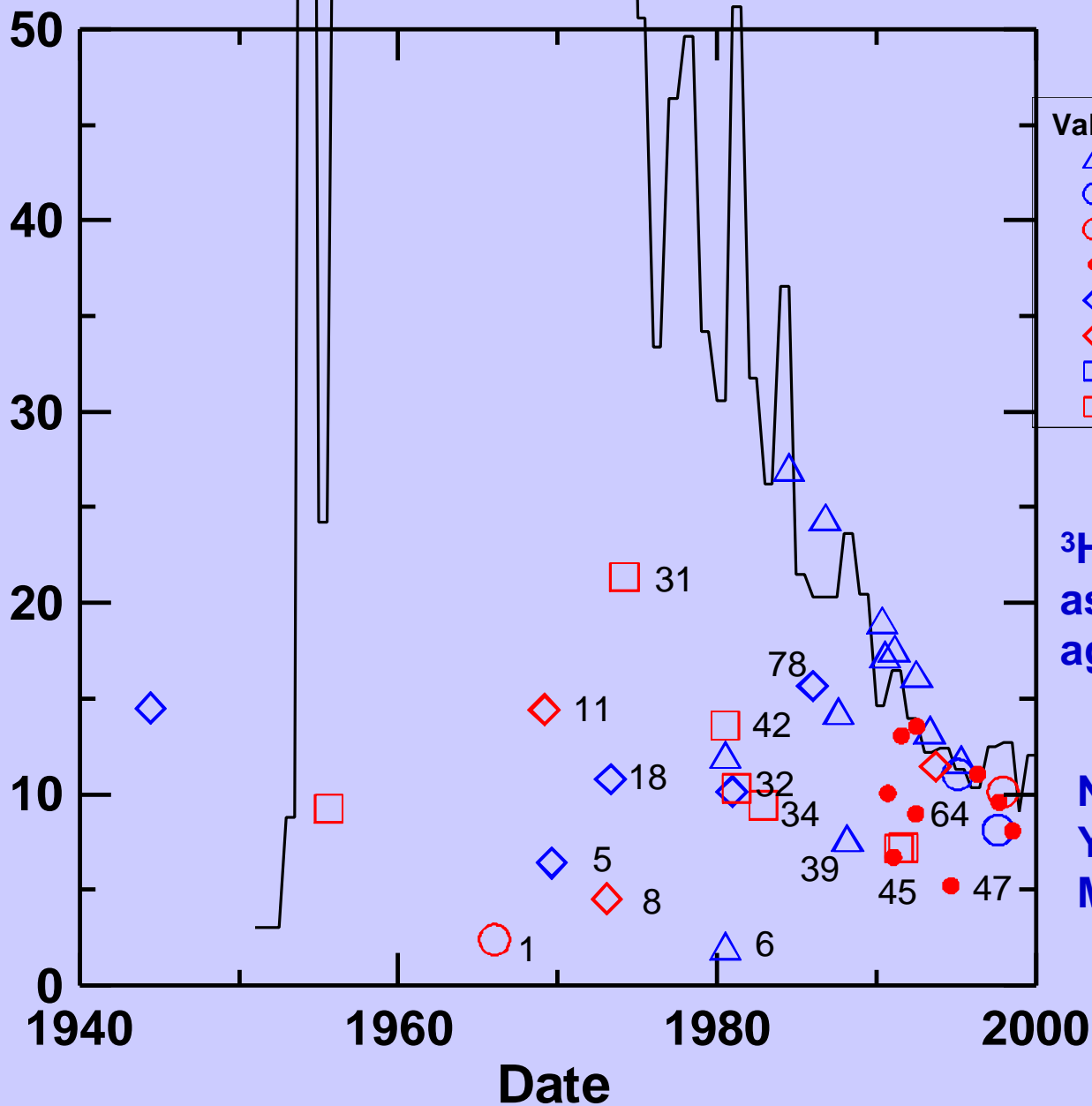


**Ages and mixing
fractions determined
from CFC-113/CFC-12**

**5-22 years and 10-100%
young water in mixture.**

**Some are inconsistent
with ^3H (CFC contam.)**

Tritium in Precipitation in TU



Valley and Ridge Carbonates

- C-Bay Sprs.
- Muddy Creek Sprs.
- Muddy Creek Dom. wells
- Muddy Creek Mon. wells
- Shenandoah Sprs.
- Shenandoah wells
- VAS Sprs.
- VAS wells

$^3\text{H} + ^3\text{He}(\text{tritogenic})$
as function of $^3\text{H}/^3\text{He}$
age.

Numbers are % of
Young fraction in
Mixture.

Bear Lithia Spring (9/2/99)

Piston-Flow Ages

CFC-11 27.2 yrs

CFC-12 27.2 yrs

CFC-113 >Modern (100 pptv)

$^3\text{H}/^3\text{He}$ 30.0 yrs

Agreement in ages suggests
Piston flow (30 yrs in pipe flow).

Major Contradiction:

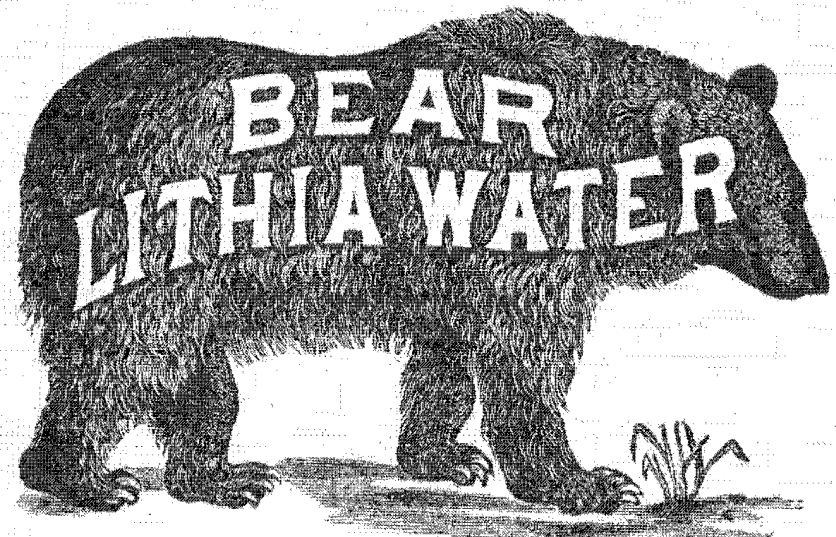
Tritium = 1.2 ± 0.2 TU

1969.6 water would contained
about 130 TU; decays to 25 TU



A. B. C. Expectorant warranted purely vegetable—no narcotics.

NATURE'S OWN REMEDY!



Cures KIDNEY and BLADDER Troubles,
URIC ACID, GOUT, and RHEUMATISM,
PHOSPHORIC DEPOSITS,
INFLAMMATION OF THE BLADDER,
DROPSICAL AFFECTIONS,
BRICK-DUST DEPOSITS,

And all forms of DYSPEPSIA which
Arise from a Non-Assimilation of Food,
Such as INDIGESTION, HEARTBURN,
FLATULENCE, SICK HEADACHE,
And a Sense of Fullness After Eating.

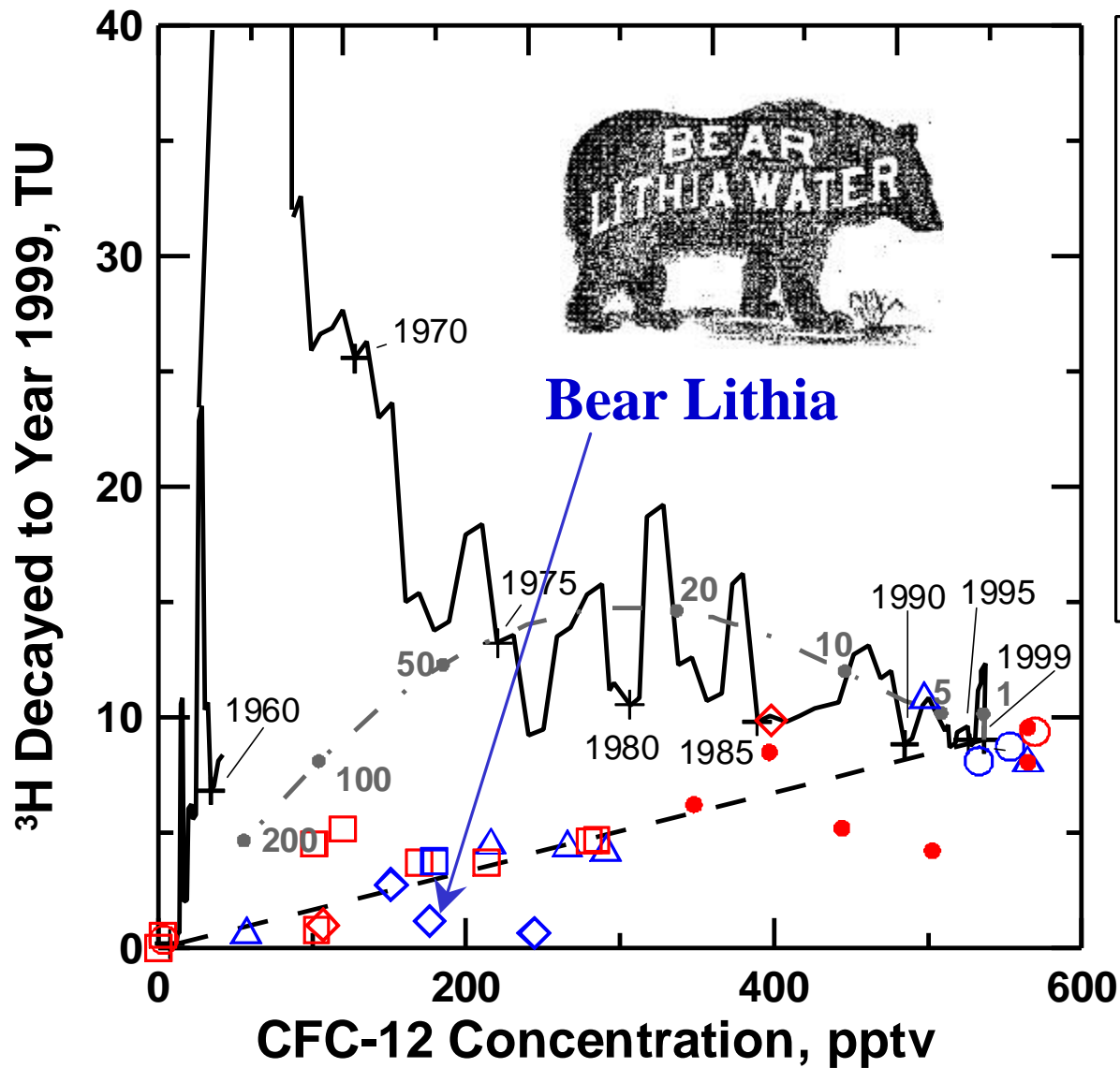
For further particulars, address

BEAR LITHIA WATER COMPANY,
ELKTON, VIRGINIA.

Price, \$3.50 per Case of one dozen half-gallon bottles, f. o. b., at
Elkton Depot, S. V. R. R., Rockingham county, Va.

AGENTS.—Polk Müller & Co., Richmond; N. Wyatt & Bro.,
Staunton; E. P. Mertz, Washington, D. C.; P. Schever & Co.,
New York; W. H. Douglass, Brooklyn, N. Y.

Routt's Emulsion is used and prescribed by the best physicians.

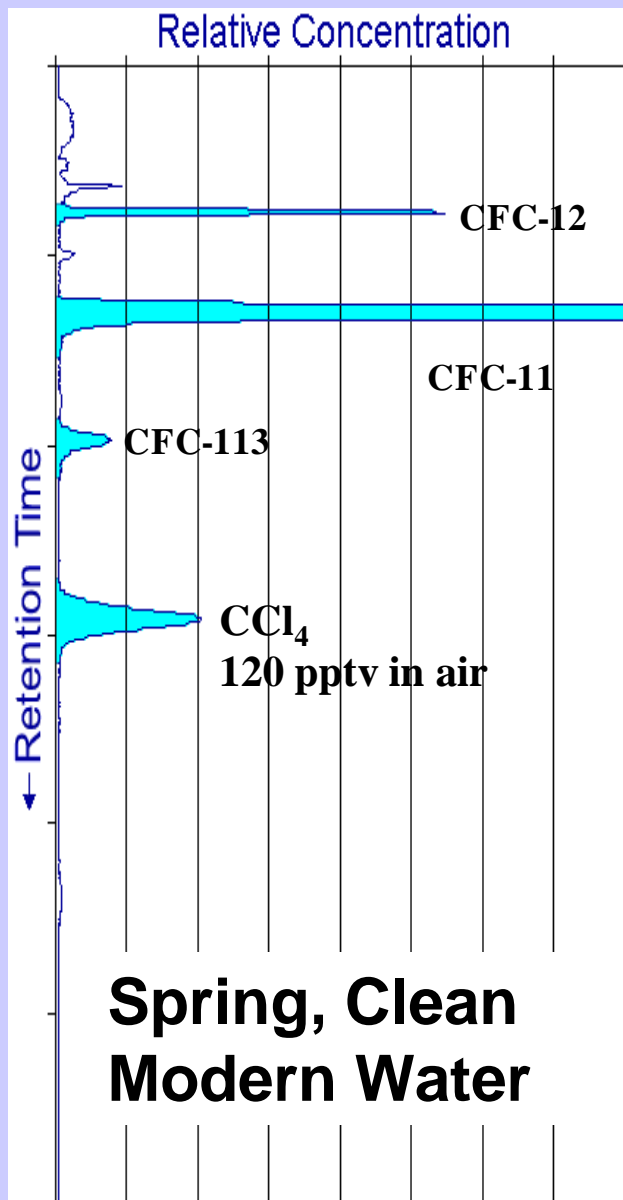


Without multiple Tracers, we would not know this is a Mixture.

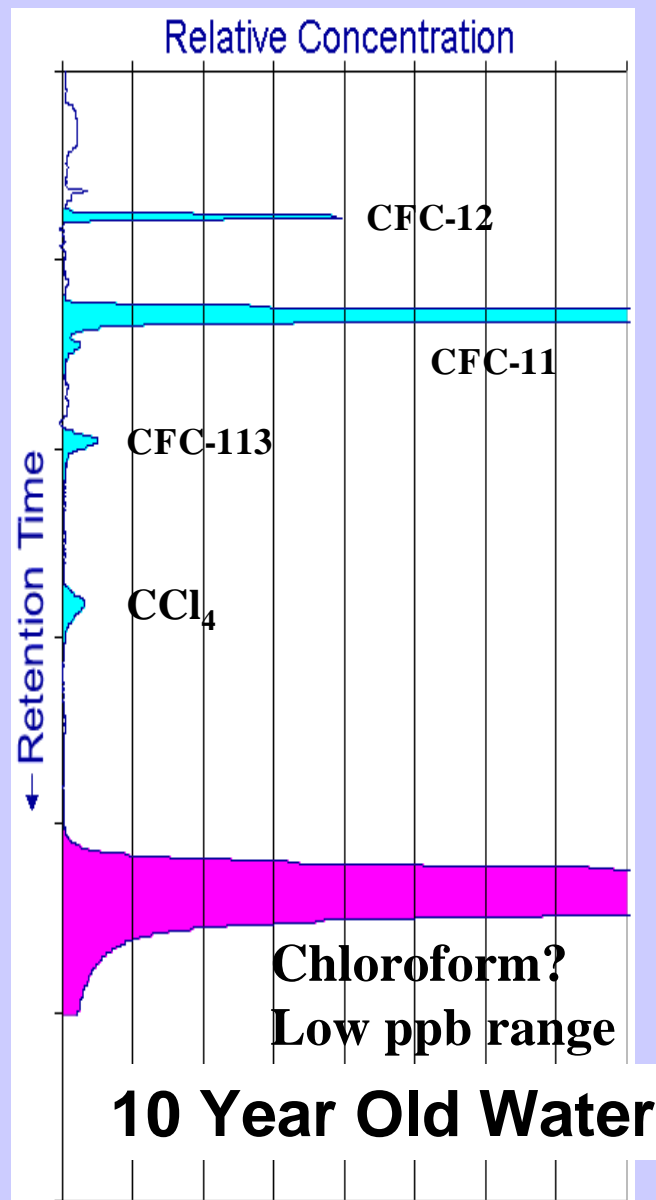
GC-ECD Chromatograms

- Analysis by GC-ECD: Purge and trap gas chromatography with electron-capture detector.
- Detects halogenated VOCs (examples: CFCs, CCl_4 , Halons, TCE, TCA, etc).
- Traditional analysis: GC-MS $< 0.1 \text{ ug/L}$ ($= 100 \text{ ng/L} = 100,000 \text{ pg/L}$)
- GC-ECD: $< 1 \text{ pg/L}$ ($= 5$ orders of magnitude below normal reporting levels)

**Shenandoah National Park
Spring
Drinking Water
Modern Clean Water**



**Spring
Yorktown, VA
Drinking Water**



A Town well in VA

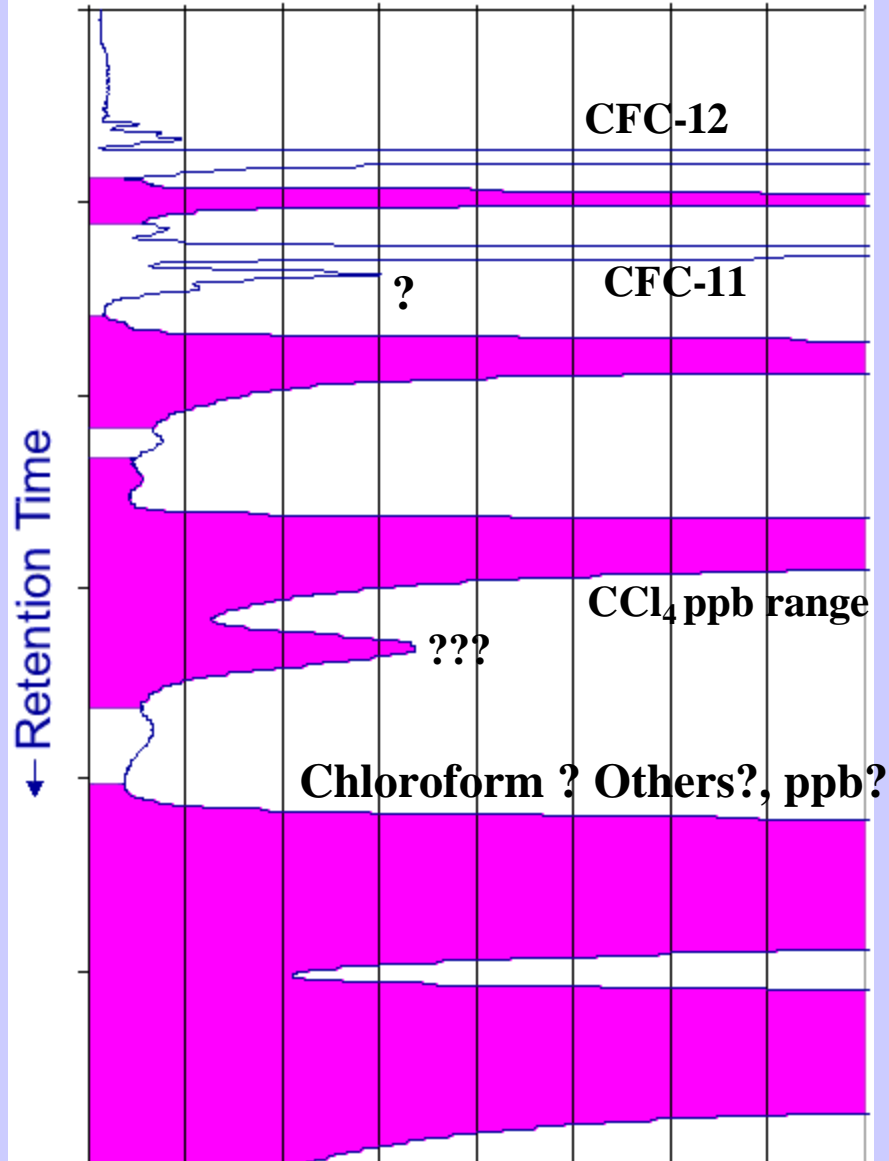
Drinking Water Well

Depth: 190 meters

Valley and Ridge

Vulnerable, fingerprints
of sources of contamination

Relative Concentration



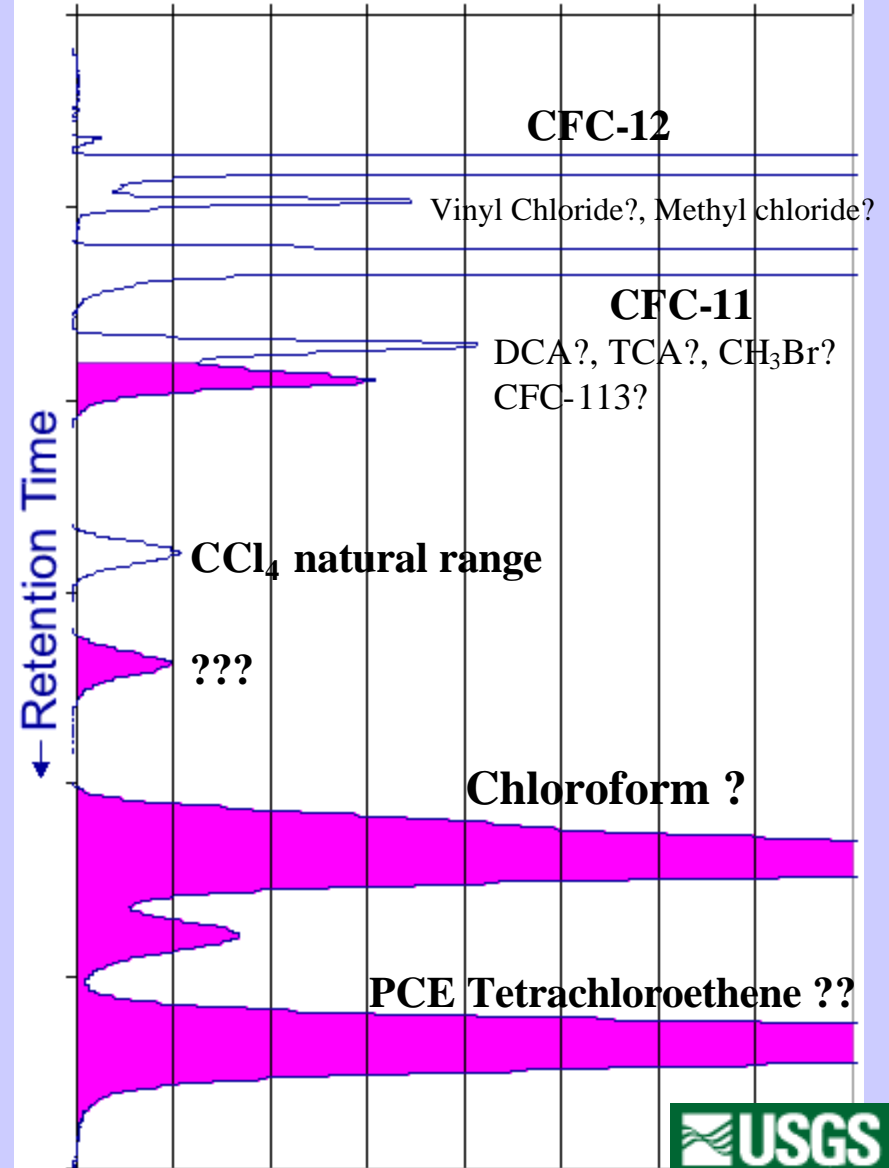
A Town well in VA

Drinking Water Well

Depth: 115 meters

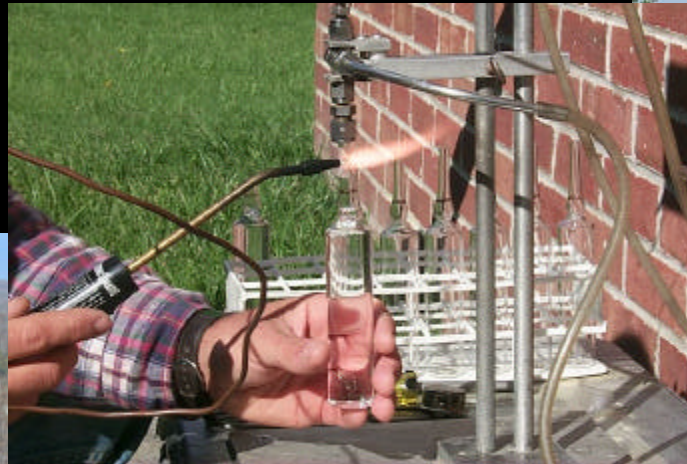
Blue Ridge

Relative Concentration



Aquifer Susceptibility in Virginia, 1998-2000

Nelms, D.L. and others,
2003, USGS
WRIR 03-4278



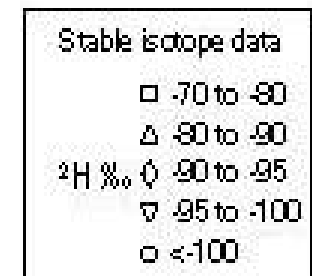
<http://water.usgs.gov/pubs/of/2003/ofr03-246/ofr03-246.htm>



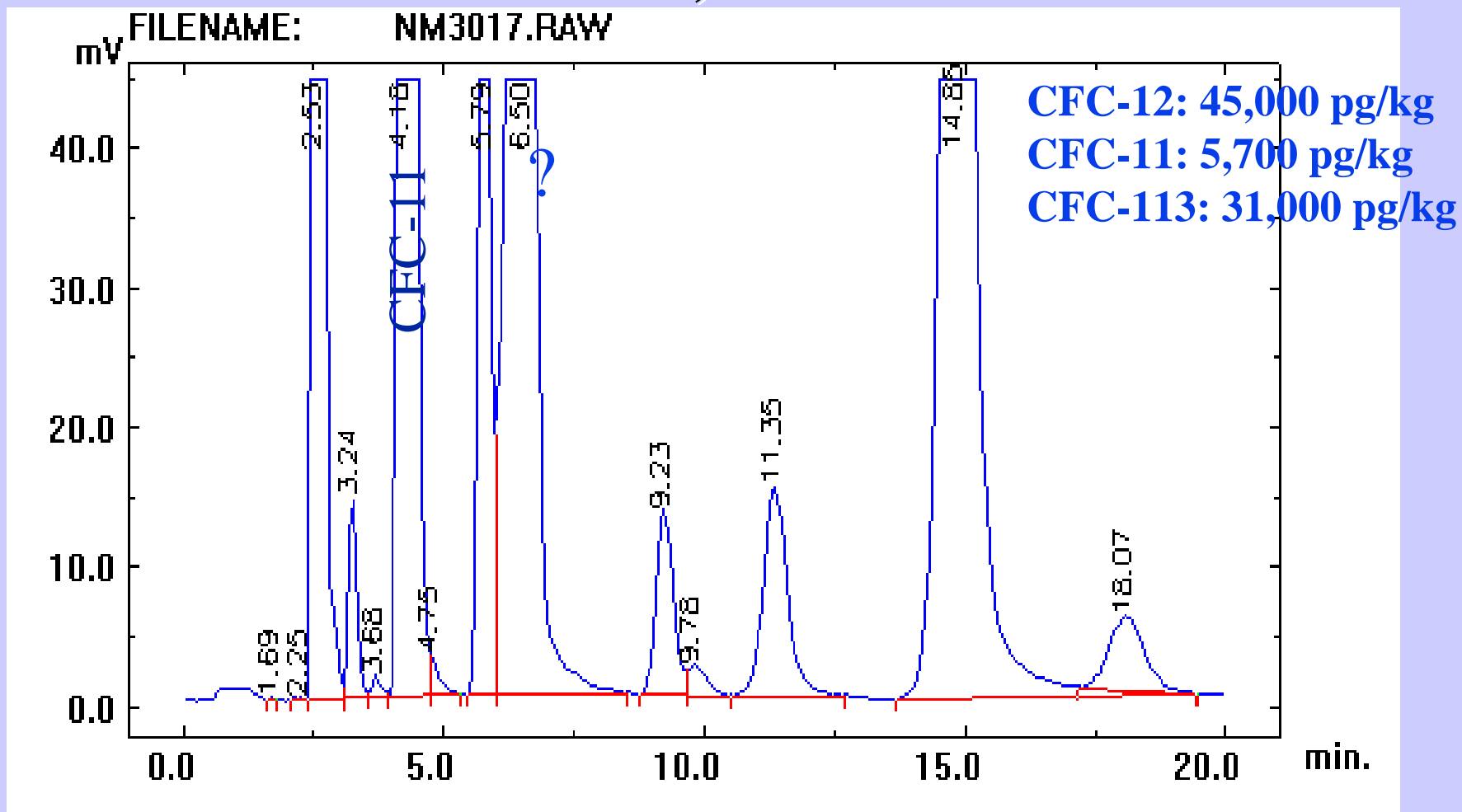
Sandia Piedmont



We'll

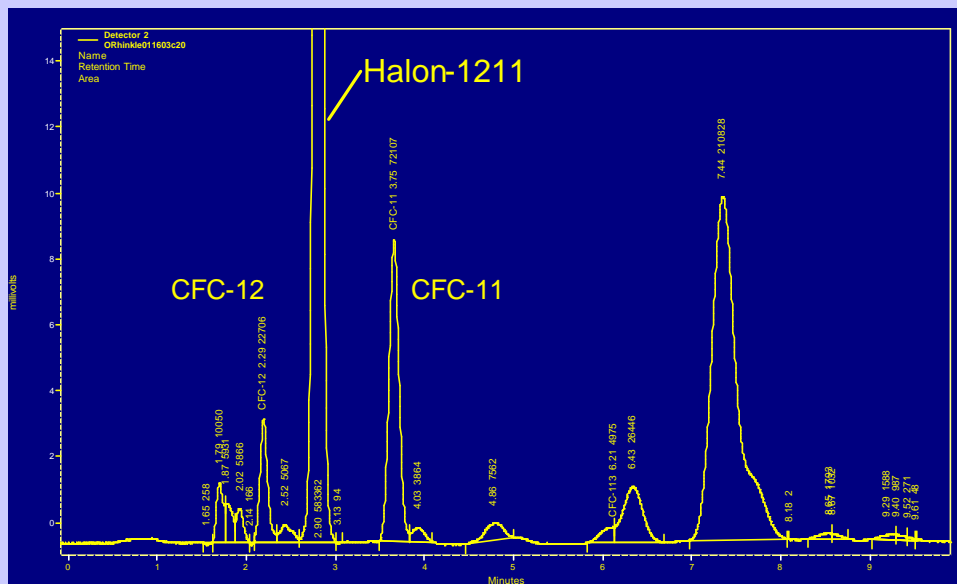


LALF-9, 6/29/96

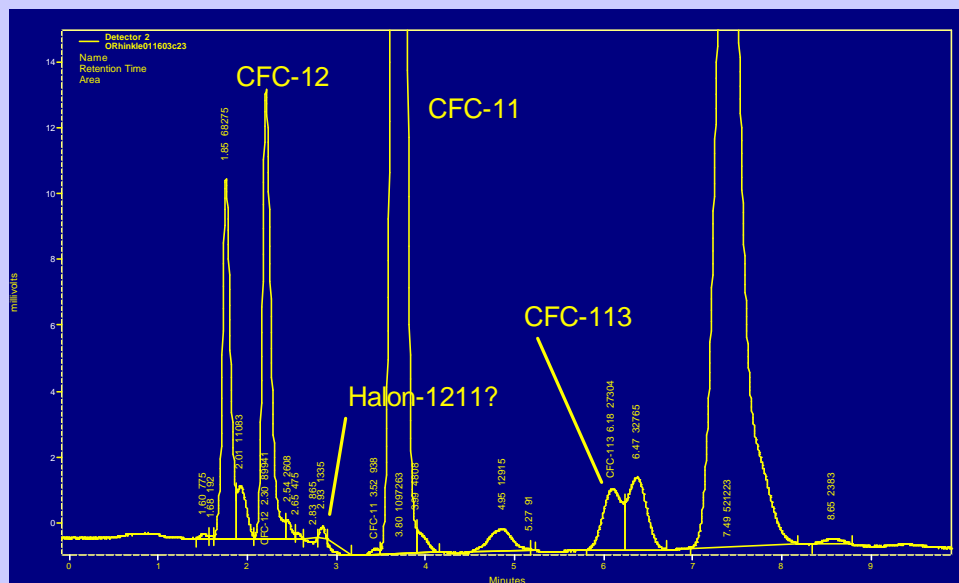


TD 233 ft, Screen 10 ft, 146 feet water above top of open interval

**Halon 1211
Drilling Tracer**



Well MW-N3D. Large Halon peak. CFC-11 and CFC-12 indicate mid- to late 1970s. Since CFCs came with Halon, water is older than 1970s, but cannot be dated further with CFCs because of contamination with drilling air. Possible mixture of old water and water contaminated with drilling air. Without the Halon data, we would have interpreted a CFC age that is too young.



Well MW-N4D. Trace or no Halon present. CFC data probably unaffected by drilling air. CFC data suggest a modern age for the water. Without the Halon data, we would not know that the CFCs are valid in this sample.

CONCLUSIONS

- Dating with environmental tracers can help answer the “When?” in forensic hydrology.
- Use of multiple tracers and “tracer plots” can help to eliminate some mixing models, and refine estimates of mean tracer age.
- To a first approximation, the ages and mixing fractions of many samples from karst or fractured rock can be interpreted using a simple binary mixing model.
- Young ages in Blue Ridge. Mixtures from wells.
- About half of the $^3\text{H}/^3\text{He}$ samples from the Valley and Ridge karst have initial tritium consistent with piston flow (0-15 yrs, unmixed). Rest are mixtures of 0-25 yrs (apparent age) mixed with old (pre-bomb) water.

CONCLUSIONS (cont.)

- Dating with ratios can be very useful. Demonstrate cases of piston flow and binary mixing, but can be affected by contamination. ($^3\text{H}/\text{CFC-12}$, $\text{CFC-113}/\text{CFC-12}$, $\text{SF}_6/\text{CFC-12}$)
- Most ground water from fractured rock or karstic aquifers is vulnerable to contamination.
- Should include tracers in well drilling.
- Use of patterns in low-level VOC detections to identify and trace sources.

CONCLUSIONS (cont.)

“... the concept of groundwater age has little significance” (Fontes, 1983).

Investigation of multiple environmental tracers in groundwater systems can often help to refine interpretation of age, and refine conceptualization of ground-water flow.

Thanks

- **Chesapeake Bay Study.** Scott Phillips, Bruce Lindsey, Gary Spieran, Mike Focazio, J.K. Bohlke, Bill Burton, Colleen Donnelly, Ed Busenberg.
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- **Noble Gas Laboratory of Lamont-Doherty Earth Observatory, Columbia University.**