

Hydrologic Benchmark Network Stations in the Midwestern U.S. 1963-95 (USGS Circular 1173-B)

Abstract and Map	List of all HBN	Introduction to	Analytical
Index	Stations	<u>Circular</u>	Methods

Blue Beaver Creek near Cache, Oklahoma (Station 07311200)

This report details one of the approximately 50 stations in the Hydrologic Benchmark Network (HBN) described in the four-volume U.S. Geological Survey Circular 1173. The suggested citation for the information on this page is:

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All of the tables and figures are numbered as they appear in each circular. Use the navigation bar above to view the abstract, introduction and methods for the entire circular, as well as a map and list of all of the HBN sites. Use the table of contents below to view the information on this particular station.

Table of Contents

- 1. Site Characteristics and Land Use
- 2. Historical Water Quality Data and Time-Series Trends
- 3. Synoptic Water Quality Data
- 4. References and Appendices

Site Characteristics and Land Use

The Blue Beaver Creek HBN Basin is in the Wichita Mountains section of the Central Lowland physiographic province in southwestern Oklahoma (Figure 20. *Map showing study area in the Blue Beaver Creek Basin and photograph of the main stream channel below Lake Jed Johnson*). The 64-km² basin ranges in elevation from 370 to 732

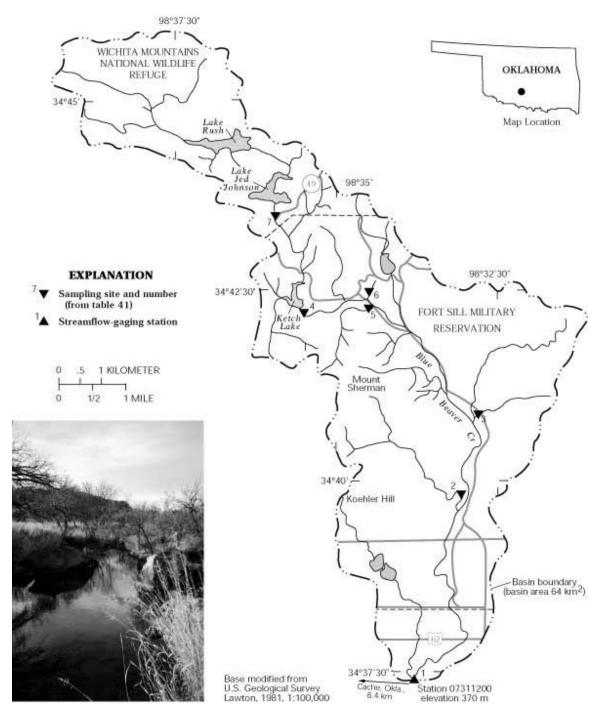


Figure 20. Map showing study area in the Blue Beaver Creek Basin and photograph of the main stream channel below Lake Jed Johnson

m and drains a landscape of smooth rounded hills flanked by a broad, gently rolling plain. The USGS gaging station is located 6.4 km east of the town of Cache, Okla., at latitude $34\times37'24"$ and longitude $98\times33'48"$. Blue Beaver Creek flows south into Cache Creek, a tributary of the Red River, and has a channel length of about 26 km upstream from the gaging station and an average stream gradient of 6.9 m/km. Mean monthly discharge varied from $0.04m^3$ /s in July to $1.1 m^3$ /s in May, and annual runoff from the basin was 19 cm from 1964 through 1995 (Blazs and others, 1996). In most years, there are periods of no flow at the gaging station, particularly during July through October. The climate is temperate continental with long, hot summers and short, mild winters. Average monthly air temperatures range from $3.0\times$ C in January to $27.8\times$ C in July (National Climatic Data Center, 1997). Precipitation averages 77 cm annually and is greatest in May (13.1 cm) and least in January (2.9 cm). Snowfall averages 12 cm annually and normally melts within a few days of falling.

The Blue Beaver Creek Basin is in the Great Plains Steppe and Shrub ecoregion (Bailey and others, 1994) and is covered by short- and mixed-grass prairies with oak savanna communities usually along the drainages. Common grass species include bluestem, indian grass, switchgrass, blue grama, and side-oats grama, and the predominant woody species are live oak, post oak, blackjack oak, and bigtooth maple. Most soils in the basin are classified as Mollisols and are mapped in the Lawton soil series (Mobley, 1967). The Lawton series consists of deep, loamy soils that formed in old alluvium of mixed rock type derived from the Wichita Mountains. A typical profile has a surface layer of brown loam to a depth of 35 cm overlying a subsoil of reddish-brown clay loam and yellowish-red sandy clay loam that extends to a depth of 150 cm. In most places, the soils have granitic sand and pebbles throughout the profile. The Stony Rock Land-Cobble Rock Land association is mapped on the slopes of the Wichita Mountains and includes a combination of exposed bedrock, shallow soils over bedrock, and cobble land, which consists of deep reddish-brown loams with between 25 and 70 percent cobblestones (Mobley, 1967).

The headwaters of Blue Beaver Creek originate in the Wichita Mountains, which are a series of igneous knobs of Cambrian age that protrude through sedimentary rocks of Permian age (Gilbert, 1982). The Cambrian rocks in the basin belong to the Carlton Rhyolite Group, which includes highly silicified rhyolitic lavas, consisting of phenocrysts of perthite and quartz in a groundmass of quartz and orthoclase. The main channel of Blue Beaver Creek cuts across the Post Oak Conglomerate and Garber Sandstone of Permian age, which were derived from sediment and debris eroded from the core of the Wichita Mountains (Havens, 1983). The Post Oak Conglomerate contains rounded boulders and cobbles of igneous rock types and fragments of limestone, dolomite, and shale eroded from older sedimentary rocks. Farther from the mountain flanks, the Post Oak Conglomerate becomes increasingly finer and eventually grades into red-brown, fine-grained sandstones and mudstones typical of the Garber Sandstone (Havens, 1983). Recent alluvium consisting of unconsolidated sand, silt, and clay was deposited along the flood plain. This alluvium ranges from 1 to 15 m in thickness and is the main source of water for domestic wells and stock supplies in the Wichita Mountains (Havens, 1983).

The Blue Beaver Creek Basin drains the western part of Comanche County in Oklahoma, and all land in the basin is publicly owned. The lower 75 percent of the basin is in the Fort Sill Military Reservation and the upper 25 percent is in the Wichita Mountains Wildlife Refuge. Land in the refuge in managed by the U.S. Fish and Wildlife Service to provide prairie grassland habitat for herds of buffalo, longhorn cattle, elk, and deer. The part of the wildlife refuge north of Lake Rush is set aside for wildlife use and is closed to the general public. Visitor use is concentrated near Lake Rush and Lake Jed Johnson where fishing and small motorized boats are permitted. These two small reservoirs and Ketch Lake to the south have a combined surface area of 53 ha. Land in the military reservation is used by the U.S. Army for tactical maneuvers (moving heavy equipment across the landscape) and foot patrols. Fort Sill uses standard erosion control and revegetation techniques to minimize the effects of intensive military training on the environment (Toni Hodgkins, U.S. Department of the Army, written commun., 1996). Since 1990, several roads in the basin have been improved, and more than 15 stream crossings have been constructed using limestone riprap to reduce erosion (Toni Hodgkins, written commun., 1996).

Historical Water-Quality Data and Time-Series Trends

The data set for the Blue Beaver Creek HBN Station analyzed for this report includes 138 water- quality samples that were collected from April 1967 through July 1995. Samples were collected monthly from 1968 through 1973 and quarterly from 1971 through 1995. Samples in the early part of the period of record were probably analyzed at a USGS laboratory in Oklahoma City, Okla., that operated until 1973 (Durum, 1978). After establishment of the central laboratory system, samples were analyzed at a laboratory in Salt Lake City, Utah, from 1973 through 1975 and at the NWQL in Arvada, Colo., from 1976 through 1995. Daily discharge records for Blue Beaver Creek (station 07311200) are available beginning in July 1964.

Calculated ion balances for 120 samples with complete major-ion analyses are shown in Figures 21a and 21b. *Temporal variation of discharge, field pH, major dissolved constituents, and ion balance at Blue Beaver Creek, Oklahoma.* Ion balances ranged from -6.4 to 16 percent, and 95 percent of samples had values within the \pm 5 percent range, indicating that the analytical results are of high quality. The average ion balance was -0.1 percent, indicating that unmeasured constituents, such as organic anions, do not seem to contribute substantially to the ion balance of stream water at this site. Time- series plots of the major dissolved constituents were inspected for evidence of method-related effects (fig. 21). The only notable pattern was the presence of several elevated potassium concentrations in the mid-1970's. Although there are no documented changes in field or laboratory methods during this period, the possibility of a methodrelated bias cannot be completely ruled out.

Median concentrations and ranges of major constituents in stream water at the Blue Beaver Creek Station and VWM concentrations in wet-only deposition measured at the Great Plains Apiaries NADP Station are listed in table 38. Precipitation chemistry at the NADP station, which is about 100 km northeast of the HBN station, is dilute and slightly acidic with a VWM pH of 5.0 for 13 years of record. The predominant cations were calcium and ammonium, which accounted for 31 and 29 percent of the total cation charge, respectively. Sulfate and nitrate were the predominant anions, contributing 53 and 35 percent of the total anion charge, respectively. The predominance of strong acid anions indicates that precipitation at the NADP station may be affected by anthropogenic emissions of sulfur and nitrogen compounds, which cause acid rain.

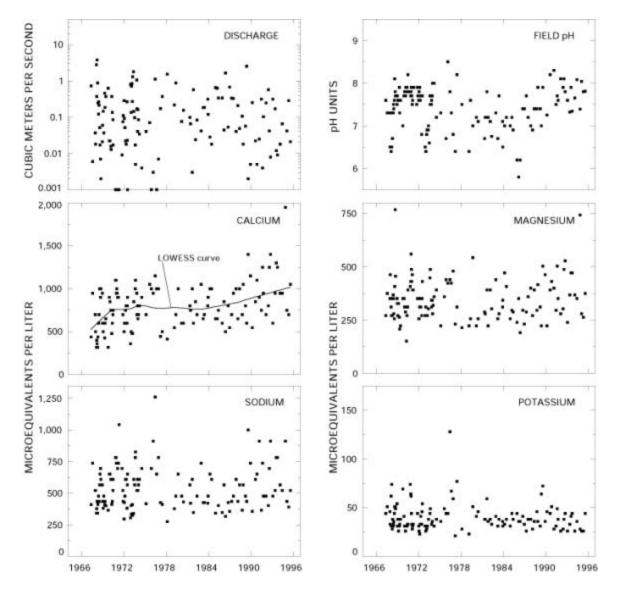


Figure 21a. Temporal variation of discharge, field pH, major dissolved constituents, and ion balance at Blue Beaver Creek, Oklahoma

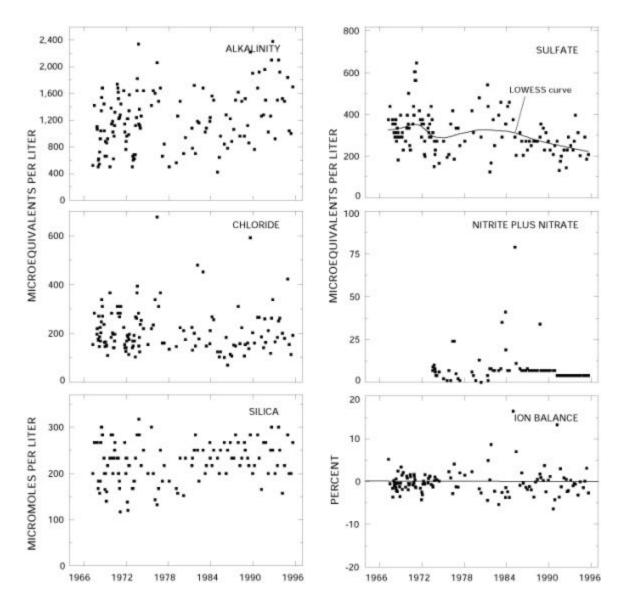


Figure 21b. Temporal variation of discharge, field pH, major dissolved constituents, and ion balance at Blue Beaver Creek, Oklahoma - Continued

Table 38. Minimum, first quartile, median, third quartile, and maximum values of physical properties and major dissolved constituents measured in water-quality samples from Blue Beaver Creek, Oklahoma, April 1967 through July 1995, and volume-weighted mean concentrations in wet precipitation collected at the Great Plains Apiaries Station, Oklahoma

[Concentrations in units of microequivalents per liter, discharge in cubic meters per second, specific conductance in microsiemens per centimeter at 25 degrees Celsius, and silica in micromoles per liter; n, number of stream samples; VWM, volume-weighted mean; inst., instantaneous; spec. cond., specific conductance; <, less than, --, not reported]

		Precipitation					
Parameter	Minimum	First quartile Median Third quartile Maximum				n	VWM ^a
Discharge, inst.	0.0006	0.018	0.075 0.28		3.8	136	
Spec. cond., field	89 130 160 220		220	400 13		10	
pH, field	5.8	7.0	7.5	7.8	8.6	135	5.0 ^b
Calcium	320	600	600 750 950		1,950	131	15
Magnesium	150	i0 270 340 400		400	770	131	2.8
Sodium	270	420	520	650	1,260	134	5.0
Potassium	20	20 31 37 44		44	130	127	1.5
Ammonium	<.7	<1.1	2.1	4.3	15	53	14
Alkalinity, laboratory	420	880	1,170	1,510	2,380	137	
Sulfate	130	250	290	350	650	135	23
Chloride	70	150	190	260	260 680		5.3
Nitrite plus nitrate	<.7	3.6	7.1	7.3	78	75	15°
Silica	120	200	230	270	320	128	

^a Data are volume-weighted mean concentrations for 1983—95.

^b Laboratory pH.

^c Nitrate only.

Stream water in Blue Beaver Creek is moderately concentrated and well buffered; specific conductance ranged from 89 to 400 mS/cm, and alkalinity was between 420 and 2,380 meg/L (table 38). The major cation in stream water was calcium, and the predominant anion was bicarbonate. The predominance of these solutes in stream water is attributed to the weathering of calcareous rock fragments in the Permian conglomerates and sandstones. The median concentration of chloride in stream water was 190 meq/L, which is more than 30 times greater than the VWM concentration of 5.3 meq/L in precipitation. On the basis of the difference between annual precipitation and runoff, evapotranspiration can account for about a fourfold increase in the chloride concentration of precipitation, indicating that most chloride in stream water is derived from sources other than precipitation. Because the basin is undeveloped, the only reasonable source of additional stream-water chloride is dissolution of salts in the underlying conglomerates. The median concentration of sulfate in stream water was 290 meq/L compared to 23 meq/L in precipitation, indicating that a substantial amount of stream-water sulfate probably is derived from weathering of the sedimentary bedrock. Concentrations of inorganic nitrogen species in stream water were generally less than the VWM concentrations in precipitation, indicating that most atmospheric nitrogen is retained by vegetation and soils in the basin.

The solute composition of stream water was further evaluated by analyzing correlations between solutes and stream discharge (table 39). The weathering-derived constituents had strong inverse correlations with discharge, except for silica (rho = - 0.180). These results are consistent with a hydrologic system where weathering-enriched base flow is diluted by water from shallow or surficial sources during periods of increased discharge. Strong positive correlations were found among calcium, magnesium, and alkalinity (0.911 < rho < 0.946), which is consistent with weathering of carbonate minerals in the Permian conglomerates. The strong positive correlation between sodium and chloride (rho = 0.847) may indicate that the primary source of chloride is weathering of halite in the sedimentary bedrock. Sulfate and silica were not correlated with discharge or other dissolved constituents, despite the fact that both solutes are presumably derived from weathering.

Table 39. Spearman rank correlation coefficients (rho values) showing the relation among discharge, pH, and major dissolved constituents, Blue Beaver Creek, Oklahoma, October 1967—95

	Q	pH	Ca	Mg	Na	K	Alk	SO ₄	Cl
pH	-0.347								
Ca	827	0.397							
Mg	837	.353	0.962						
Na	825	.339	.854	0.868					
K	204	051	.211	.212	0.250				
Alk	783	.469	.945	.912	.843	0.229			
SO ₄	176	386	.052	.077	.205	.004	-0.086		
Cl	715	.124	.666	.682	.847	.204	.671	0.322	
Si	180	.130	.397	.379	.103	.386	.386	174	0.097

[Q, discharge; Ca, calcium; Mg, magnesium; Na, sodium; K, potassium; Alk, alkalinity; SO₄, sulfate; Cl, chloride; N, nitrite plus nitrate; Si, silica]

Results of the seasonal Kendall test for trends in discharge and major dissolved constituents are listed in table 40. Statistically significant trends were detected in the unadjusted calcium and sulfate concentrations at the 0.01 probability level. Results were similar for the flow-adjusted concentrations, indicating that trends in calcium and sulfate were not driven by variations in stream discharge. After flow adjustment, a statistically significant upward trend also was detected in stream-water alkalinity. The LOWESS curves in figure 21 show that the downward trend in sulfate and the upward trend in calcium are primarily due to changes in concentrations since the mid-1980's. Although not shown in figure 21, the pattern in flow-adjusted alkalinity was similar to the pattern for unadjusted calcium concentrations. Trends in stream chemistry at the Blue Beaver Creek Station may reflect changes in environmental factors, such as precipitation chemistry or land-use activities. Lynch and others (1995) reported a decline in the sulfate concentration of precipitation at a nearby NADP station in Arkansas from 1982 through 1990. Although the direction of the trend in sulfate deposition is consistent with the stream-water trend, the precipitation trend [-0.3 (meq/L)/yr] is an order of magnitude smaller than the stream-water trend [-3.2 (meq/L)/yr], indicating that atmospheric deposition alone cannot account for the trend in stream-water sulfate. The large increases in stream-water calcium [9.5 (meq/L)/vr] and alkalinity [14.1 (meq/L)/vr] also cannot be accounted for by trends in atmospheric deposition and may be linked to land-use activities in the basin. Military training exercises, particularly those involving the

movement of heavy vehicles, have caused serious land damage and have caused increased sedimentation to some streams on the military reservation (Toni Hodgkins, U.S. Department of the Army, written commun., 1996). Since 1990, however, Fort Sill has implemented techniques to control erosion, such as revegetation, road improvements, and construction of hardened stream crossings. Although most of these measures might be expected to decrease land- use effects on streams, the hardened crossings are constructed of limestone riprap, which could provide a readily soluble source of dissolved calcium and alkalinity to the stream.

Table 40. Results of the seasonal Kendall test for trends in discharge andunadjusted and flow-adjusted pH and major dissolved constituents, Blue BeaverCreek, Oklahoma, April 1967 through July 1995

[Trends in units of microequivalents per liter per year, except discharge in cubic meters per second per year, pH in standard units per year, and silica in micromoles per liter per year; inst., instantaneous --, not calculated]

Parameter	Unad	justed	Flow adjusted			
T dramotor	Trend	p-value	Trend	p-value		
Discharge, inst.	< 0.01	0.634				
pH, field	.01	.230	0.01	0.066		
Calcium	7.1	.009	9.5	.000		
Magnesium	<.1	.957	1.1	.175		
Sodium	<.1	.706	1.1	.325		
Potassium	2	.037	2	.247		
Alkalinity, laboratory	7.2	.088	14.1	.000		
Sulfate	-3.5	.002	-3.2	.004		
Chloride	-1.5	.097	7	.365		
Nitrite plus nitrate	<.1 ^a	.759				
Silica	<.1	.880	<.1	.999		

^a Trend calculated for 1973—95 using a trend test for censored data.

Synoptic Water-Quality Data

Results of the surface-water synoptic sampling in the Blue Beaver Creek Basin on February 2 and 3, 1991, are listed in table 41, and the locations of the sampling sites are shown in figure 20. During the sampling period, discharge at the gaging station was about 0.07 m^3 /s compared to the median daily discharge of about 0.06 m^3 /s for February (Lawrence, 1987), indicating that the basin was sampled during normal flow conditions for that time of year. The concentrations of most solutes measured at site 1 during the synoptic sampling were similar to the first-quartile concentrations reported for the HBN station during the entire period of record (table 38). The composition of the tributary streams was similar to stream water collected at site 1, with calcium and sodium the dominant cations and bicarbonate and sulfate the dominant anions. Ion balances for all the synoptic samples were positive (range 0.4 to 6.1 percent), indicating that unmeasured constituents, such as organic anions, may have been an important component of stream water during the sampling period. Stream-water chemistry in the basin was quite variable, particularly for the weathering-derived constituents. For example, calcium ranged from a minimum of 140 meg/L at site 5 to 650 meg/L at site 1 and alkalinity, which ranged from 140 meq/L at site 5 to 950 meq/L at site 1. This pattern in stream chemistry seems to reflect the distribution of bedrock types in the basin. Sites with higher concentrations (sites 1 and 3) drain areas underlain by the Permian conglomerates, whereas the more dilute sites (sites 2, 4, 5, and 7) drain large areas of Carlton Rhyolite. Spatial patterns in sulfate and chloride concentrations also seemed to be related to bedrock type; higher concentrations were at sites 1 and 3, and lower concentrations were at sites 2, 4, 5, and 7. Site 6 was the one tributary that did not fit the general pattern by having a low alkalinity similar to streams draining igneous rocks, but having elevated sulfate and chloride similar to streams draining sedimentary rocks. The low silica concentration at site 4 compared to site 7 was somewhat unexpected and may indicate silica removal by diatoms in Ketch Lake. The effect of military activities on water quality in the basin was difficult to evaluate because the sedimentary rocks underlie the same part of the basin that is used for military training. Therefore, it was not possible to determine if differences in water quality between site 7 (upstream from military activities) and site 1 (basin outlet) predominantly reflected the influence of geology or the effect of military activities in the basin.

Table 41. Physical properties and major dissolved constituents from surface-water sampling sites in the Blue Beaver Creek Basin, Oklahoma, collected February 2—3, 1991

[Site locations shown in fig. 20; Q, discharge in cubic meters per second; SC, specific conductance in microsiemens per centimeter at 25 degrees Celsius; pH in standard units; Ca, calcium; Mg, magnesium; Na, sodium; K, potassium; Alk, alkalinity; SO₄, sulfate; Cl, chloride; NO₃, nitrate; Si, silica; concentrations in microequivalents per liter, except silica in micromoles per liter; <, less than; --, not reported; criteria used in selection of sampling sites: BG = bedrock geology, LU = land use]

Site	Station number	Q	SC	pН	Ca	Mg	Na	K	Alk	SO_4	Cl	NO_3	Si	Criteria	Remarks
1	7311200	0.065	130	7.8	650	260	390	22	950	230	120	<4.1	180	BG, LU	Downstream from Fort Sill
2	343946098330200		77	6.9	320	170	250	17	500	170	70	.7	230	BG	Mixed lithology
3	344050098325000	.0003	120	6.9	420	190	480	22	700	250	140	<5.5	250	BG	Mixed lithology
4	344214098353300	.017	71	7.8	350	150	200	31	430	140	70	1.4	80	BG, LU	Rhyolite, below reservoirs
5	344216098342700		38	6.6	140	74	150	9	140	180	42	.7	250	BG	Rhyolite
6	344224098342700		77	7.1	340	160	220	13	230	310	140	1.4	170	BG	Rhyolite
7	344327098355700	.0042	65	7.3	320	140	170	26	450	110	48	2.1	140	BG, LU	Rhyolite, upstream from Fort Sill

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Appendix A. List of Map References

- a. U.S. Geological Survey topographic maps:
 - Meers, Oklahoma (1:24,000), 1975
 - Mount Scott, Oklahoma (1:24,000), 1975
 - Quanah, Oklahoma (1:24,000), 1975
 - Saddle Mountain, Oklahoma (1:24,000), 1956
 - Taupa, Oklahoma (1:24,000), 1970, gaging station on this quadrangle
 - Lawton, Oklahoma (1:100,000), 1981
- b. Geologic maps:
 - Hoffman, M.G., 1930, Geology and petrology of the Wichita Mountains: Oklahoma Geological Survey Bulletin 52, 83 p., scale 1:63,360.
- c. Soil surveys:
 - Mobley, H.L., 1967, Soil survey, Comanche County, Oklahoma: U.S. Department of Agriculture Soil Conservation Service, 58 p.
- d. Miscellaneous maps:
 - Havens, J.S., 1977, Reconnaissance of the water resources of the Lawton quadrangle, southwestern Oklahoma: Oklahoma Geological Survey Hydrologic Atlas HA–6, 4 sheets, scale 1:250,000.
 - Havens, J.S., 1983, Reconnaissance of groundwater in vicinity of Wichita Mountains, southwestern Oklahoma: Oklahoma Geological Survey Circular 85, 13 p., sheet 1 of 2, geologic map, scale 1:250,000.

Appendix B. NWIS Site-Identification Numbers

Table B-1. NWIS identification numbers and names for water-quality sampling sites.

Site	Identification Number	Site Name
1	07311200	BLUE BEAVER CR NR CACHE OKLAHOMA
2	343946098330200	KOEHLER HILL TRIB NR CACHE OKLAHOMA
3	344050098325000	NEWT JONES HILL TRIB NR MEDICINE PK OKLAHOMA
4	344214098353300	BLUE BEAVER CR BELOW KETCH L NR THE HOLY CITY
		OKLAHOMA
5	344216098342700	MT SHERMAN TRIB NR THE HOLY CITY OKLAHOMA
6	344224098342700	KETCH RD TRIB NR THE HOLY CITY OKLAHOMA
7	344327098355700	BLUE BEAVER CR BELOW JED JOHNSON L OKLAHOMA